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TRANSMISSION OF DOPPLER BROADENED RESONANCE RADIATION
THROUGH ABSORBING MEDIA WITH COMBINED

DOPPLER AND PRESSURE BROADENING
[NITRIC OXIDE Y-BANDS AS AN EXAMPLE]

ENGINE TEST FACILITY

ARNOLD ENGINEERING DEVELOPMENT CENTER

AIR FORCE SYSTEMS COMMAND

ARNOLD AIR FORCE STATION, TENNESSEE 37389

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20. ABSTRACT (Continued) value of C was found to be 1270 $^{\pm}$ 200 K/atm. This value of C leads to a value for the optical diameter of NO for broadening of 3.5 $^{\pm}$ 0.3 Å. The determination of the broadening parameter permits the accurate calculation of the transmission of NO γ -band radiation through the high temperature, ambient pressure media corresponding to jet engine exhausts, in turn making possible the relating of transmission measurements to the concentration of NO. The application of the measurement and calculation procedure to the measurement problem is discussed.

PREFACE

The research reported herein was conducted by the Arnold Engineering Development Center (AEDC), Air Force Systems Command (AFSC), under Program Element 65807F. The results were obtained by ARO, Inc. (a subsidiary of Sverdrup & Parcel and Associates, Inc.), contract operator of AEDC, AFSC, Arnold Air Force Station, Tennessee. The work was accomplished in the Engine Test Facility (ETF), under ARO Project Numbers R32P-54 and R32P-55A. The authors of this report were M. G. Davis, W. K. McGregor, J. D. Few, and H. N. Glassman, ARO, Inc. The manuscript (ARO Control No. ARO-ETF-TR-75-115) was submitted for publication on June 30, 1975.

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1.0 INTRODUCTION

The radiative transfer of Doppler broadened spectral lines, contained within electronic-vibrational-rotational bands, through Doppler broadened absorbing media has been treated in a previous report (Ref. 1). In the work reported herein, the radiative transfer of Doppler broadened source bands through media with Doppler and collision broadened lines is treated. In Ref. 1, the nitric oxide (NO) molecule was used to demonstrate the technique and to compare experimental data from low-pressure, Doppler broadened absorbing media with the calculated band profiles. In the work reported herein, NO will be used again but at pressures such that collisional broadening by a foreign gas (N2) is present. Knowledge of the collisional broadening parameter for NO was found to be uncertain (Ref. 2) so that a major part of the present work was the determination of the broadening parameter to a higher degree of accuracy over a large range of pressure and temperature conditions.

The application of this work is found in the use of spectral line transmission through high temperature media to determine species concentration and temperature in situ. Previously, measurements using this spectral line resonance absorption technique (Refs. 3 and 4) have been made using empirical procedures to account for the finite width of the source lines. The work reported in Ref. 1 permitted the measured transmission to be related to species concentration and temperature in a rigorous fashion when pressure (e.g., collisional) broadening could be neglected. The development in the present report will permit measurements to be related to species concentration and temperature over pressure ranges where the pressure broadening is comparable to, or dominates, the Doppler broadening.

The broadening parameter (a') as used in most treatments of spectral line broadening (Refs. 5 and 6, for example) is proportional to the ratio of the sum of the natural half-width and the collisional half-width of the line to the Doppler half-width. The natural half-width can be neglected for electronic resonance transitions. Theoretical estimating procedures are generally inadequate to predict the collisional broadening and measurements cannot separate the Doppler from the collisional contributions so that the parameter (a') must be experimentally determined. Direct measurement of the broadening half-width to determine a' requires high resolution

spectral instruments. In the treatment used in the present study, a theoretical calculation procedure is used to predict transmission from experimental measurements, and thus by iteration of the calculations using a' as a parameter, the value of a' is indirectly determined by matching calculated and measured transmission. The required experimental data were obtained by measurement of the transmission of Doppler broadened lines in the (0,0) γ -band of NO from a gas discharge lamp through a temperature-controlled (60 to 1,000°F) absorption cell containing known mixtures of NO and N₂ at pressures varying from about 0.1 to 2 atm.

The results of the theoretical development and the empirically determined value of a' are applicable directly to calculations of the transmission of the (0,0) γ -band of NO emitted by a resonance radiation source through absorbing media of various NO concentrations, pressures, and temperatures. The temperatures and pressures accessible in the calibration laboratory is limited, so that direct calibration for all possible conditions is not possible. However, the determination of a' and the use of the theoretical relationships between the transmissivity and NO concentration permits extension to environments expected at the exhaust exit of combustion engines and other devices in which measurements of NO concentration might be desired.

2.0 THEORETICAL DEVELOPMENT

2.1 DEVELOPMENT OF TRANSMISSION FORMULAS FOR CASES INVOLVING ABSORPTION LINES BROADENED BY FOREIGN GASES

For a single, isolated j^{th} spectral line, the transmission $(\overline{T_j})$ of a source line having some frequency distribution $(I_{\nu_j}^o)$ through uniform absorbing medium of length ℓ is given by (Ref. 5):

$$\overline{T}_{j} = \int_{0}^{\infty} I_{\nu_{j}}^{o} \exp\left(-k_{\nu_{j}} \ell\right) d\nu \tag{1}$$

where ν is the frequency and $k_{\nu j}$ is the absorption coefficient which has a frequency distribution independent of $I_{\nu j}$. If the radiation source is maintained at low pressure, the frequency distribution for $I_{\nu j}$ can be attributed to the Doppler effect, and is given by

$$I_{\nu_{j}}^{\circ} = I_{\nu_{j}}^{\circ} \exp \left\{ -\left[\frac{2(\nu - \nu_{j}^{\circ})}{(\Delta_{s}\nu_{j})_{D}} \sqrt{\underline{\ell_{n}} 2} \right]^{2} \right\}$$
 (2)

where $I_{\nu_0}^{\bullet}$ is the intensity of the source line at center frequency (ν_j^{\bullet}) and $(\Delta_s \nu_j)_D$ is the Doppler width at half the intensity (half-width) of the emitted spectral line. The Doppler half-width of the source line is given by

$$(\Delta_s \nu_j)_D = 2 \nu_j^o \sqrt{\frac{2 \ell_{n,2} \kappa T_s}{M_s c^2}}$$
 (3)

In Eq. 3, κ is Boltzmann's constant, T_S is the absolute temperature of the source, M_S is the molecular weight of the emitting molecule, and c is the speed of light.

In general, the frequency distribution of the absorption coefficient (k_{ν_i}) is given by (Ref. 6):

$$k_{\nu_{j}} = k_{\nu_{j}^{0}} \frac{1}{\pi} \int_{-\infty}^{\infty} \frac{a'e^{-y^{2}}}{a'^{2} + (\omega_{j} - y)^{2}} dy$$
 (4)

where

$$a' = \frac{(\Delta_a \nu_j)_L}{(\Delta_a \nu_j)_D} \sqrt{\ell_n 2}$$
 (5)

$$\omega_{j} = \frac{2(\nu_{j} - \nu_{j}^{\diamond})}{(\Delta_{a}\nu_{j})_{D}} \sqrt{\ell_{n} 2}$$
(6)

and y is a dummy variable of integration. In Eq. 4, $k_{\nu_j^0}$ is the absorption coefficient at line center for Doppler conditions. In Eq. 5, $(\Delta_a \nu_j)_L$ is the Lorentz half-width (due to collision broadening) of the absorption line and is given by (Ref. 6):

$$(\Delta_{a}\nu_{j})_{L} = \frac{Z_{L}}{\pi} \tag{7}$$

where Z_L is the frequency of collisions between the absorbing molecules and the surrounding molecules which leads to broadening of the energy states of the absorbing molecules. The factor $(\Delta_a \nu_j)_D$ is the Doppler half-width of the absorption line and is given by

$$(\Delta_a \nu_j)_D = 2 \nu_j^{\circ} \sqrt{\frac{2 \ell_{n2} \kappa T_a}{M_a c^2}}$$
 (8)

where T_a is the static temperature of the absorbing medium and M_a is the molecular weight of the absorbing molecules.

Equation 4 for $k_{\nu_{\dot{1}}}$ can be shown to reduce to (Ref. 7):

$$k_{\nu_{j}} = k_{\nu_{j}^{\circ}} R[\exp[-(\omega_{j} + ia')^{2}] \operatorname{erfc}[-i\omega_{j} + a']]$$
(9)

where R denotes the real part and $i = \sqrt{-1}$. It should be noted that, for low pressures and high temperatures, a' is very small and Eq. (9) reduces to

$$k_{\nu_i} = k_{\nu_i^{\circ}} e^{-\omega^2} \tag{10}$$

This is the Doppler case examined in Ref. 1.

Close examination of Eq. (9) shows that the absorption line half-width increases and the absorption coefficient at line center decreases as a increases, as illustrated graphically in Fig. 1.

Equations (1), (2), and (4) can be combined to give the transmission of a single line through a medium:

$$T_{j} = I_{\nu_{j}^{o}}^{o} \int_{0}^{\infty} \exp \left\{ -\left[\frac{2(\nu - \nu_{j}^{o})}{(\Delta_{s}\nu_{j})_{D}} \sqrt{\ln 2} \right]^{2} \right\} \exp \left\{ -\ell k_{\nu_{j}^{o}} \frac{1}{\pi} \int_{-\infty}^{\infty} \frac{a'e^{-y^{2}}dy}{a'^{2} + (\omega_{j} - y)^{2}} \right\} d\nu (11)$$

If there are other absorption lines which might contribute to the measured transmission of the jth line, Eq. (4) must be replaced by

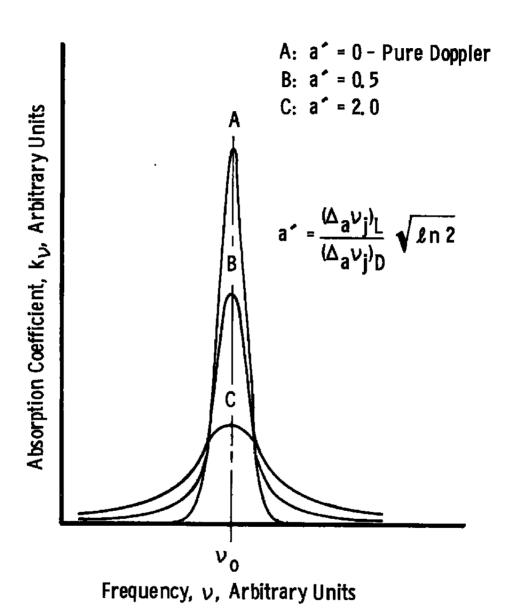


Figure 1. Illustration of spectral line shape for different values of the ratio of the pressure broadened half-width to the Doppler half-width.

$$k_{\nu} = \frac{1}{\pi} \sum_{i} k_{\nu_{i}^{c}} \int_{-\infty}^{\infty} \frac{a' e^{-y^{2}}}{a'^{2} + (\omega_{i} - y)^{2}} dy$$
 (12)

where the summation is over all absorption lines, including the jth line, which have a finite value of the absorption coefficient contributing to absorption of the jth emission line. The transmission of the radiation in the jth emission line due to absorption by many lines is given by

$$\overline{T}_{j} = I_{\nu_{j}^{\circ}}^{\circ} \int_{0}^{\infty} \exp \left\{ -\left[\frac{2(\nu - \nu_{j}^{\circ})}{(\Delta_{s}\nu_{j})_{D}} \sqrt{\ell_{n} 2} \right]^{2} \right\} \exp \left\{ \frac{\ell}{\pi} \sum_{i} k_{\nu_{i}^{\circ}} \int_{-\infty}^{\infty} \frac{a'e^{-y^{2}} dy}{a'^{2} + (\omega_{j} - y)^{2}} \right\} d\nu \quad (13)$$

The total transmission (\overline{T}) in a particular frequency interval (e.g., bandpass of a spectrometer or interference filter) is given by summing over all the emission lines in that interval and results in the equation:

$$\overline{T}_{\Delta\nu} = \sum_{j} I_{\nu_{j}^{\circ}}^{\circ} \int_{0}^{\infty} \exp \left\{ - \left[\frac{2(\nu - \nu_{j}^{\circ})}{\Delta_{s}\nu_{j}^{\circ}} \sqrt{\ell_{n} 2} \right]^{2} \right\} \exp \left\{ - \frac{\ell}{\pi} \sum_{i} k_{\nu_{i}^{\circ}} \int_{-\infty}^{\infty} \frac{a'e^{-y^{2}} dy}{a'^{2} + (\omega_{i} - y)^{2}} \right\} d\nu$$
 (14)

where the summation over j includes all emission lines which have components falling within the frequency interval of interest $(\Delta \nu)$. The transmissivity (t) or fractional transmission in a particular frequency interval $(\Delta \nu)$ is then given by

$$t_{\Delta\nu} = \frac{\sum_{j} I_{\nu_{j}^{\circ}}^{\circ} \int_{0}^{\infty} exp \left\{ -\left[\frac{2(\nu - \nu_{j}^{\circ})}{(\Delta_{s}\nu_{j})D} \sqrt{\ell n \ 2} \right]^{2} \right\} exp \left\{ -\frac{\ell}{\pi} \sum_{i} k_{\nu_{i}^{\circ}} \int_{-\infty}^{\infty} \frac{a'e^{-y^{2}} dy}{a'^{2} + (\omega_{i} - y)^{2}} \right\} d\nu}{\sum_{j} I_{\nu_{j}^{\circ}}^{\circ} \int_{0}^{\infty} exp \left\{ -\left[\frac{2(\nu - \nu_{j})}{\Delta_{s}\nu_{j}^{\circ}} \right] \sqrt{\ell n \ 2} \right]^{2} \right\} d\nu}$$
(15)

In order to carry out the evaluation of $t_{\Delta\nu}$, values of k_{ν} , and a must be determined for a particular medium and path.

2.2 RELATIONS BETWEEN ABSORPTION COEFFICIENT AND ABSORBING MEDIUM PROPERTIES

The absorption coefficient of a diatomic molecule for Doppler conditions at line center frequency (k_{ν}) is given by (Ref. 5):

$$k_{\nu_{i}^{0}} = \frac{2e^{2}\sqrt{\pi \ell n^{2}}}{me^{2}} \frac{N_{J}^{n} f_{J}^{n}}{(\Delta_{a}\nu_{i})_{D}}$$
(16)

where e is the charge on an electron, c is the velocity of light, m is the mass of an electron, $f_{J'J'}$ is the oscillator strength of the appropriate absorption line, $N_{J''}$ is the number density of molecules in the lower energy state of the molecule corresponding to the ith line, J'' is the rotational quantum number of the lower energy state, and J' is the rotational quantum number of the upper energy state.

The number density of the lower energy state (N_J ") under equilibrium conditions is given by

$$N_{J''} = \frac{hc B_o(2J''+1) exp \left[-\frac{hc}{\kappa T_a} F(J'')\right]}{2 \kappa T_a} N_o$$
 (17)

where N_0 is the number density of the molecule of interest, B_0 is the rotational constant for the ground state, h is Planck's constant, κ is Boltzmann's constant, and F(J'') is the rotational energy term for the lower rotational energy state.

The value for the oscillator strength $(f_{J'J'})$ is given by (Ref. 8):

$$f_{J'J''} = f_{v'v''} \frac{\nu_{J'J''}}{\nu_{v'v''}} \frac{S_{J''J'}}{2(2J''+1)(2S+1)}$$
 (18)

where $f_{v'v}$, is the band oscillator strength, $\nu_{J'J}$, is the frequency of the line of interest, $\nu_{v'v}$, is the frequency of the band head, $\delta_{J''J'}$ is the normalized Hönl-London factor for the line of interest, and S is the spin quantum number. Combining Eqs. (16), (17), and (18) gives

$$k_{\nu_{i}^{c}} = \frac{e^{2}\sqrt{\pi \ell n 2} h B_{o \nu_{J'J''} f_{v'v''}} S_{J''J'} N_{o} \exp \left[-\frac{hc}{\kappa T_{a}} F(J'')\right]}{2(2S+1) mc^{2} \kappa T_{a} \nu_{v'v''} (\Delta_{a} \nu_{i})_{D}}$$
(19)

For a particular spectral line of a given molecular species for which the various molecular parameters in Eq. (19) are known, values of $k_{\nu_i^{\bullet}}$ can be calculated as functions of N_o and T_a .

For the (0,0) γ -band of NO, Eq. (19) reduces to (Ref. 1):

$$k_{\nu_{1}^{0}} = 1.603 \times 10^{-14} \frac{\delta_{J'J''N_{0}}}{T_{a}^{3/2}} \exp \left[-1.4383 \,\mathrm{F}(J'')/T_{a}\right]$$
 (20)

where the cgs system of units is used throughout

2.3 EXAMINATION OF THE BROADENING PARAMETER (a')

Equation (7) states that collisional broadening of a spectral absorption line depends on the collisional frequency of the absorbing molecules with the surrounding molecules. It can be shown from classical kinetic theory that the broadening collisional frequency (Z_L) for an absorbing molecule of molecular weight (M_a) is given by

$$Z_{L} = \sum_{\ell} Z_{\ell} = 2 \sum_{\ell} N_{\ell} \sigma_{\ell}^{2} \sqrt{2\pi \kappa T_{a} \left(\frac{1}{M_{\ell}} + \frac{1}{M_{a}}\right)}$$
 (21)

where M_{ℓ} is the mass of the ℓ th type of colliding molecule causing the broadening. N_{ℓ} is the concentration of the ℓ th type of molecule, and $\sigma_{\ell}^{\,\,2}$ is the effective collisional cross section for the broadening process by the ℓ th type molecule.

Combining Eqs. (5), (7), (8), and (21) results in the equation,

$$\mathbf{a}' = \frac{\lambda_{j}^{o}}{\sqrt{\pi}} \sum_{\ell} N_{\ell} \sigma_{\ell}^{2} \sqrt{1 + \frac{M_{a}}{M_{\ell}}}$$
 (22)

where λ_{j}° is the wavelength of the absorption line at line center and results from the fact that

$$\lambda_{j}^{o} = \frac{c}{\nu_{i}^{o}} \tag{23}$$

By using the equation of state for a perfect gas, it can be shown that

$$N_{\ell} = 9.66 \times 10^{19} \frac{P_{\ell}}{T_{c}} \tag{24}$$

where p_{ℓ} is the partial pressure of the ℓ th type molecule in torr and T is in K. Combining Eqs. (22) and (24) results in

$$a' = \frac{5.45 \times 10^{19}}{T} \lambda_{j}^{\circ} \sum_{p} p_{\ell} \sigma_{\ell}^{2} \sqrt{1 + \frac{M_{a}}{M_{\ell}}}$$
 (25)

For many cases, the foreign gas is composed of chiefly one constituent, and the concentration of the absorbing gas is relatively small. In such cases, self broadening is negligible, and Eq. (25) reduces to

$$a' = \left[5.45 \times 10^{19} \lambda_j \sigma^2 \sqrt{1 + \frac{M_a}{M_f}}\right] \frac{P_a}{T_a}$$
 (26)

or

$$\mathbf{a'} = \mathbf{C_j} \frac{\mathbf{p_a}}{\mathbf{T_a}} \tag{27}$$

where σ^2 is the effective collisional cross section for the broadening process by the foreign gas, M_f is the mass of the foreign gas, and C_i is a constant for the jth line given by

$$C_j = 5.45 \times 10^{19} \lambda_j \sigma^2 \sqrt{1 + \frac{M_a}{M_f}}$$
 (28)

It is this constant (C_j) that must be determined experimentally.

3.0 DETERMINATION OF THE BROADENING PARAMETER (a') FOR THE NO MOLECULE IN THE PRESENCE OF N₂

The value of a is functionally dependent on pressure and temperature as shown by Eq. (27). To completely define a', the value of C_j must be determined experimentally. Although C_j depends on the wavelength of the spectral line of interest, for most spectral bands C_j may be considered a constant for all lines in the band and simply designated by C (i.e., λ_j changes by less than 0.1 percent throughout the (0,0) γ -band of NO). In this section, the experimental determination of C_j for a few individual lines of the (0,0) γ -band of NO and the determination of C for the entire, unresolved band will be described.

3.1 DESCRIPTION OF THE EXPERIMENTAL APPARATUS

The experimental apparatus consisted of a resonance gas discharge source lamp, a heated absorption cell, and two spectrometers. The arrangement of the apparatus is illustrated in Fig. 2.

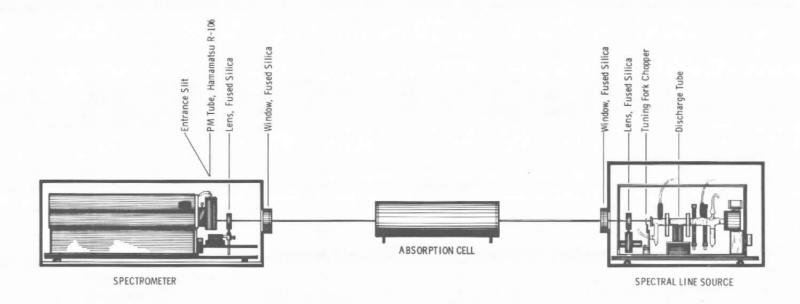


Figure 2. Schematic of experimental apparatus for resonance line absorption measurements.

3.1.1 Ultraviolet Spectrometers

A 1-m Jarrell-Ash grating spectrometer equipped with curved slits was used to obtain high resolution data. The grating has 1,180 lines/mm and was blazed for maximum reflection at 7,500 Å. All data taken with the 1-m spectrometer was in the second order of the spectrum. An RCA 1P28 photomultiplier tube with S-spectral response was used as the detector. The internal optics consisted of two fused silica lenses, each with an f-number of 4.4. The lenses were placed as shown in Fig. 2, so that parallel light was directed from the source through the absorbing gas and focused on the slit of the spectrometer. The 1-m spectrometer was operated with a physical slit width of 10 μ , resulting in an equivalent slit width of 0.03 Å, which gives sufficient resolution to separate several lines in the γ -bands of NO (Ref. 3).

A 1/2-m Jarrell-Ash grating spectrometer equipped with curved slits was used to obtain low resolution band spectra. The grating had 2, 360 lines/mm and was blazed for maximum reflection at 3,000 Å. The external optics and detector were identical to those used on the 1-m spectrometer. The 1/2-m spectrometer was operated with a physical slit width of 200 μ , resulting in an equivalent slit width of 1.6 Å. No lines of NO γ -bands could be resolved using this slit width. The 1/2-m instrument, operated with the 200- μ slit width, has been used in lieu of the 1-m instrument for field measurements of the absorption of NO γ -band radiation (Ref. 4) because it is less susceptible to vibration and misalignment problems.

3.1.2 Absorption Cell

The absorption cell used in this research study was a specially designed 91.4-cm long by 10.2-cm-diam fused-silica tube with flat fused-silica end plates and 1/2-in. tubes for gas entry and exit (Fig. 3). The tube was enclosed in a copper sleeve which was wrapped with a Calrod unit for heating purposes. A highly reflective aluminum sheet was wrapped around the heating unit, and the entire assembly was surrounded by three inches of insulation. The assembly was encased in a steel housing. Three-inch diameter holes were left in each end of the tube so that the light source could be directed through the cell and into the spectrometer.

The heating unit was connected to a temperature controller which controlled the temperature of the gas within the cell through the monitoring of strategically placed thermocouples. The temperature could

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be varied between ambient and 1,000°F and controlled within an accuracy of ±1 percent. The pressure of the gas in the cell was measured by means of pressure transducers over a range of from 50 to 1,500 torr, with an accuracy of ±1 percent.

The gas system is shown schematically in Fig. 4. The absorber gases could be admitted to the cell and sealed off by valves or could be flowed continuously. In this experiment, the gases were admitted and sealed off. The gases used were mixtures of NO in N₂ which were supplied by Scott Research Corporation as calibration gases. Three mixtures were used: (1) 40 parts per million (ppm), (2) 100 ppm, or (3) 400 ppm of NO. The actual mixtures were determined to be 40, 99, and 395 ppm, using a gas chromatograph, with an accuracy of ±5 percent.

3.1.3 Resonance Lamp Source Characteristics

A schematic of the resonance lamp used as a line source is presented in Fig. 5. The source was run using a 12:3:1 mixture (by volume) of (A:N2:O2) at approximately 5-torr pressure with an applied voltage of 2, 800 v. Radiation emitted at the end of the water-cooled capillary tube was directed through the absorption cell and into the optics of the spectrometer (Fig. 2). The gas temperature in the capillary tube was maintained at approximately 320 K by the watercooling jacket. It is assumed that the dominant broading mechanism under these conditions is due to the Doppler effect. The Doppler line width (Eq. (3)) for the lines of the (0,0) γ -band of NO at 320 K is 0.0005 Å, so that the actual width of the lines is much smaller than the equivalent slit width of the spectrometer (0.03 Å). A spectrum of the (0,0) band from the lamp in which many of the lines are resolved is given in Fig. 6. In order to employ the computational technique developed in Section 2.0 and the spectrometer simulation described in Ref. 1, it is necessary to define the relative intensity of each line in the band. By using all the resolved lines in the spectra of Fig. 6, a plot of the radiation intensity divided by the relative line strength versus the upper state energy (F_J ') was made (Fig. 7). The upper state energy parameter varies from the energy at J'=1/2 to the energy at J'=81/2. A more complete discussion of the energy levels of the upper state for the (0, 0) γ -bands appears in Ref. 3.

In order to find the relative intensity of those lines that are not resolvable, it is necessary to use the curve in Fig. 7. For a particular line, the value of $I_{J'J''}/\delta_{J'J''}$ corresponding to its value of $F_{J'}$ is found from Fig. 7 and is then multiplied by the appropriate value of $\delta_{J'J''}$ resulting in a value of the relative intensity $(I_{J'J''})$ for the J'J''

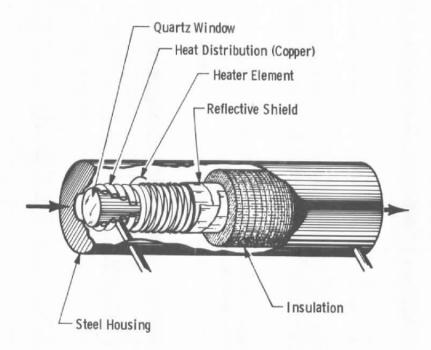


Figure 3. Diagram of heated absorption cell.

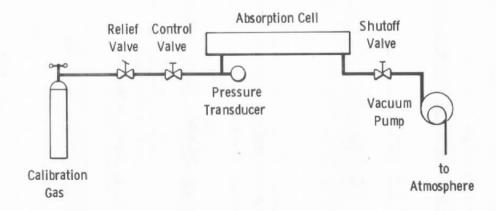


Figure 4. Schematic of gas handling system for heated absorption cell.

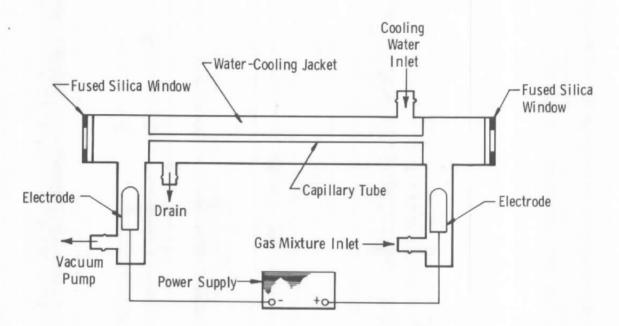


Figure 5. Diagram of resonance lamp used to produce narrow-line radiation.

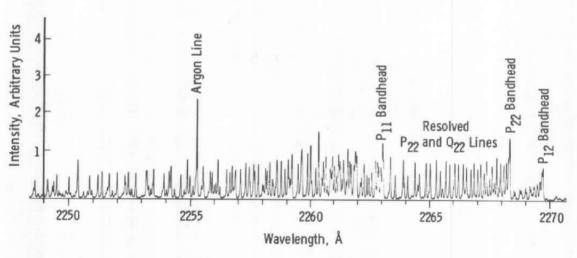


Figure 6. (0,0) band of the NO γ-system obtained from discharge tube containing a mixture of 12:3:1 (by volume) of A:N₂:O₂ at 8 torr with 2,800 v applied by use of 1-m spectrometer in second order (equivalent slit width, 0.03 Å).

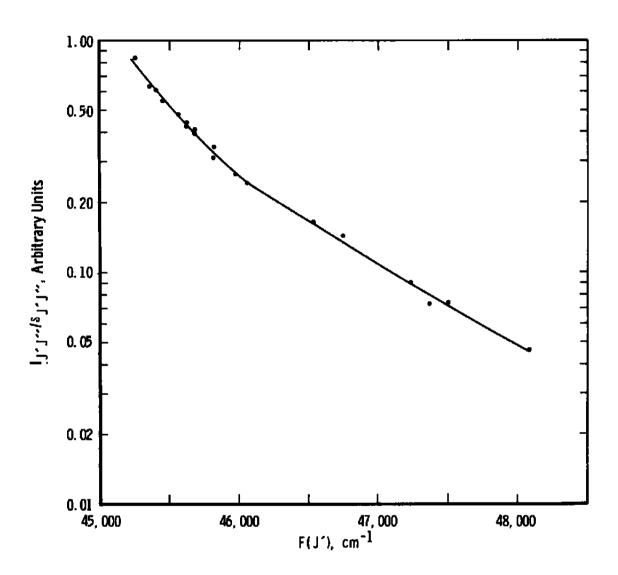


Figure 7. Population distribution of excited rotational states of the $A^2\Sigma$ level of NO in a water-cooled discharge tube operated at 8 torr with 2,800 v applied and containing 12:3:1 mixture (by volume) of A:N₂:O₂.

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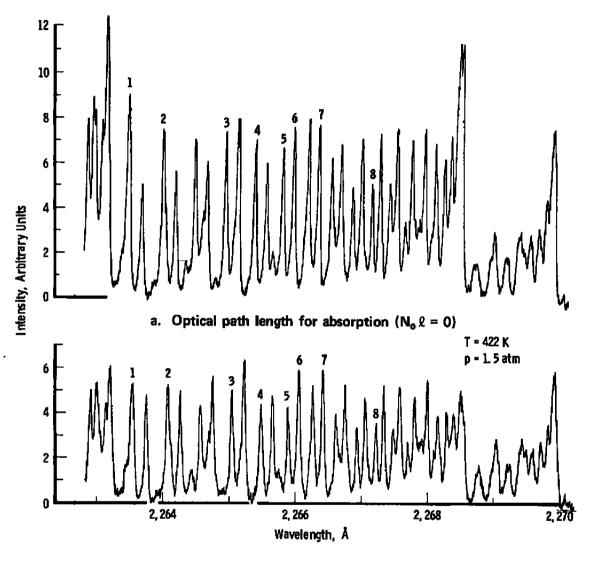
line. Values for $\delta_{J'J}$ are obtained from the formulas developed by Earls (Ref. 9) as given in Ref. 3. Experimental values of ν_j , as given by Deeszi (Ref. 10) were used in these calculations. The experimental values of ν_j are more accurate than can be calculated using the Hill and Van Fleck formula as discussed in Ref. 1. Values for I_{ν_j} used in the transmission equations (Eqs. (13), (14), or (15)) are thus calculated from the values of $I_{J'J''}/\delta_{J'J''}$ in Fig. 7.

The distributions of line intensities were measured several times during these experiments using pressures in the resonance lamp ranging from 5 to 15 torr with corresponding changes in the applied voltage and current. No measurable changes in the relative distribution of line intensities shown in Fig. 7 have been found, although the level of intensity may change considerably with the lamp operating conditions.

3.2 PROCEDURES AND RESULTS

The procedure for determining the value of C for NO consisted of the following steps:

(1) A series of laboratory spectral absorption measurements of the (0, 0) γ -band of NO were made for several partial pressures of NO and No and for several gas temperatures. Several sets of high resolution data were taken using the 1-m spectrometer to measure the spectral transmission at the various conditions. A portion of the resolved spectrum for one of these tests is shown in Fig. 8 with no gas in the cell and with a mixture of NO and No in the cell. For the case illustrated in Fig. 8, the number density of NO is 2.58×10^{15} molecules/cm³, the pressure is 1.5 atm, and the temperature is 422 K. Several lines used to obtain values of a are identified in Fig. 8. The transmissivity (t_j) for each of the lines was determined. Data comparable to that shown in Fig. 8 were obtained for three conditions in the absorption cell. Low resolution measurements were also made with the 1/2-m spectrometer in order to determine the transmission of the unresolved band. Example data for the band transmission are shown in Fig. 9 with the operating conditions stated on the figure. The transmissivity at the second band-head was determined from the data in Fig. 9. Unresolved band transmission data for the second band-head were obtained at 27



b. Optical path length for absorption ($N_o \ell = 2.36 \times 10^{17}$ molecules/cm²) Figure 8. Example of transmission measurements for NO (0,0) γ -band, as obtained with a 1-m spectrometer having an equivalent slit width of 0.03 Å.

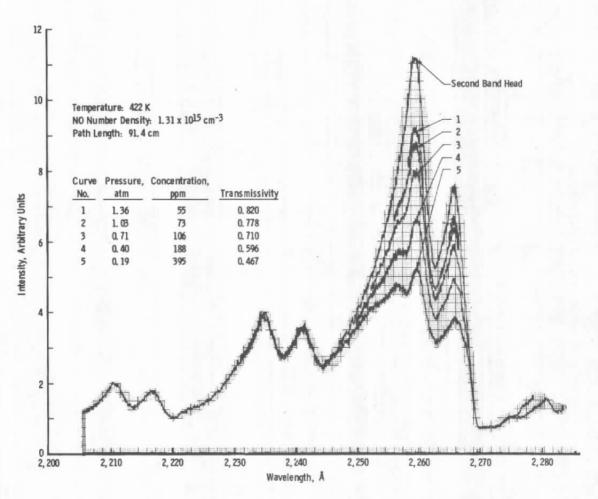


Figure 9. Example set of transmission measurements for NO (0,0) γ -band, as obtained with a 1/2-m spectrometer having an equivalent slit width of 1.6 $\,\mathring{\rm A}$.

- conditions of pressure, temperature, and NO concentration in the cell.
- (2) By using the computer simulation of spectra described in Ref. 1 and utilizing Eq. (14), a series of calculated transmitted spectra for many values of a was obtained for several values of NO concentration, temperature, and with the spectrometer slit width identical to that used in the laboratory experiments. An example synthetic spectrum for the high resolution simulation is shown in Fig. 10, and an example for the low resolution simulation is shown in Fig. 11. The computer program for spectral simulation is presented in Appendix A.
- (3) By comparing the laboratory spectra taken at a given set of pressure, temperature, and concentration conditions in the absorption cell with the computer simulated spectra, which was calculated for the same conditions and at various values of a', the value of a' which best matches the experimental result was determined. The procedure is illustrated by comparing Fig. 8 with Fig. 10 and Fig. 9 with Fig. 11. computed spectra led to values of transmission for varying values of a ', as illustrated in Fig. 12, which is taken from the second band-head transmission for various a values in Fig. 11. The values of a' in Figs. 10 and 11 were the values which gave the best match between the measured and computed spectra. The values were chosen in this way so that the reader might make a direct comparison of the simulated and measured spectra. In practice, the a'values for a particular experimental condition were obtained from plots like Fig. 12, which were obtained from simulated spectra over a range of arbitrarily selected values of a'.

The results obtained from the high resolution spectral absorption measurements are summarized in Table 1, and the results from the low resolution data are summarized in Table 2. For the high resolution data, eight individual lines were selected for determining the values of C_j . These lines are numbered in Table 2 to correspond to Figs. 8 and 10. For the low resolution data, the transmission at the second bandhead as marked in Figs. 9 and 11 was used for determining the values of a'.

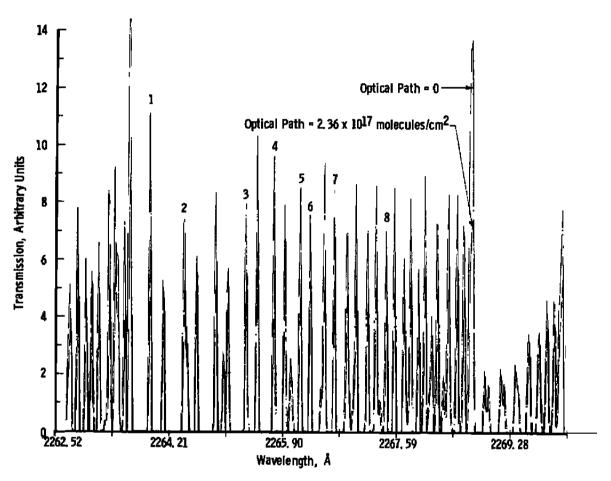


Figure 10. Example of computed transmission for NO (0,0) γ -band for an equivalent spectral slit width of 0.03 Å and at conditions corresponding to Fig. 8.

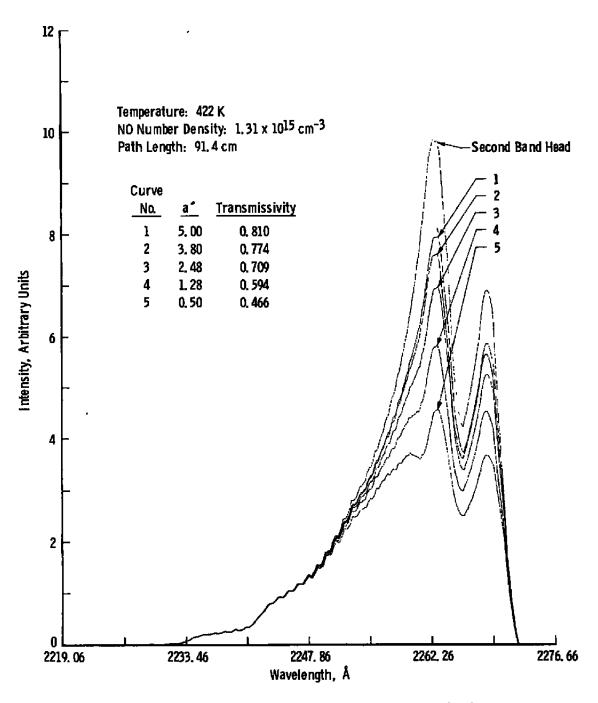


Figure 11. Example set of computed transmission for NO (0,0) γ -band for an equivalent spectral slit width of 1.6 Å and at conditions corresponding to Fig. 9.

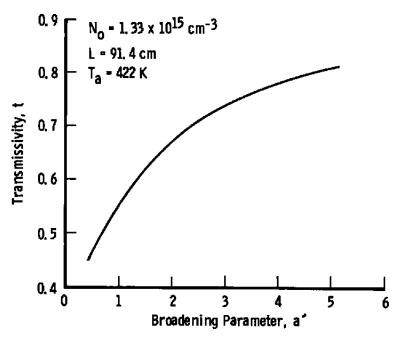


Figure 12. Calculated transmissivity (t) of the second band head of the (0,0) NO γ -band as a function of the line broadening parameter (a').

Table 1. Determination of Broadening Parameter (a') from Resolved Lines of the (0,0) Band of the NO γ -System

	Apporbing Medium Parameters				
Line Number	NO, Molecules/cm ³	Pressure,	Temperature, K	t	a'
1 '	2.58 x 10!5	1 500	422	0.84	4.4
2 1				0.73	: 4.6
3				0,70	4.
4 [1 1	0, 67	4.4
5			i	0.66	4.
6			1 1	0.85	4.1
7	1	1	'	0, 82	4,
8	▼	₹	. 🔻	0.76	4.1
1	10.32×10^{15}	1,500	422	0 19	4.
2	- 1	1	l i	0.28	4. 6
3	İ	ľ	' i	0.25	4, (
4		1	<u> </u>	0,21	4.1
5			1	0.21	4.
6	ļ	ł		0.51	4.1
7	l l	J	i []	0.45	4.8
а	*	7] + ;	0, 27	4.1
1	l x 10 ¹⁵	0. 102	296	0,65	0.1
2				0.66	0.7
3				0.54	0. 7
4			ı I I	0.52	0.5
5				0 51	0, 5
6				O. B4	0, 5
7			! [0.78	0.6
8	*	*	*	0.59	0, 7

Table 2. Determination of Broadening Parameter (a') from the Unresolved (0,0) Band of the NO γ -System

Absorbing Medium Parameters		T. Second Band		
NO, Molecules/cm ³	Pressure, atm	Temperature, K	Peak	a'
1.33 x 10 ¹⁵	0.190	422	0.465	0.50
	0.395	422	0,591	1.28
•	0.707	422	0.710	2.48
	1.034	422	0.777	3.80
7,52 x 10 ¹³	0, 721	744	0.977	1.00
	1.362	744	0, 9.88	2,65
1.36×10^{14}	0. 190	4 11	0.900	0,30
1. 30 X 10-5	0. 395	411	0, 942	1.18
	0. 707	411	0,965	2,54
	0.701	411	0,303	2, 34
3.314×10^{14}	0.190	422	0.792	0.34
	0.395	422	0.870	1.26
	0. 721	422	0.920	2.63
	1.034	422	0.936	3.70
2. 75 x 10 ¹⁵	0, 395	422	0.323	1.00
	0.707	422	0.436	1.84
	1.034	422	0,526	2.74
	1.361	422	0.586	3.55
1.61 x 10 ¹⁵	0. 408	744	0.498	0.34
	0,721	744	9, 658	1, 21
	1.034	744	0, 752	2.22
3, 91 x 10 ¹⁵	0,395	296	0, 270	1.95
U, UI A IV	0, 393	296 296	0.330	2,75
	1,020	296	0,402	3, 77
	1,350	296	0, 468	5. 18
		200	1	••••
1.89 x 10 ¹⁵	0. 190	296	0.370	1.03
	0.395	296	0. 450	1.65
	0.707	296	0.600	3, 35

The tabulated values of a 'are plotted as a function of the ratio p/T in Fig. 13, and a straight line given by a least mean square fit is drawn through the points. The value of C in Eq. (27) (shown as C_j) is equal to the slope of the line in Fig. 12. For the (0,0) γ -band of NO, C is found to be 1,270 ± 200 K/atm. From Eq. (28), the value of the optical cross section (σ^2) is found to be (9.5 ± 1.5) x 10⁻¹⁶ cm². The optical diameter is 3.5 ± 0.3 Å approximated by Thorsen and Badger in Ref. 2 from the experimental work carried out by Weber and Penner (Ref. 11).

4.0 DISCUSSION

The major use for the work reported herein is to determine calibration curves for the transduction of transmission measurements of NO γ -band radiation through media containing NO to the number density of NO in the media. Such calculations are used to extend laboratory calibrations at limited pressures and temperatures to a wider range of values; thus accurate values of the broadening parameter becomes a necessity. An application of the use of the calculations to a measurement situation is described in Ref. 12, where calibration curves are generated for absorption through the exhaust of a turbine engine burner at temperatures well above those possible in a laboratory absorption cell.

The uncertainty in values of a' projects the uncertainty in the final values of concentration determined from the γ -band absorption measurements. The estimated uncertainty in the values of a' given in Fig. 13 is about ± 16 percent. When this value of uncertainty is used to determine the density of absorbers, the projected uncertainty in the density is about ± 10 percent and constitutes the principal uncertainty in the resonance absorption technique for concentration measurements through uniform media. A more accurate value of a' would correspondingly reduce the possible uncertainty in measuring the concentration of NO. However, experimental errors in measurements of this nature would probably limit the uncertainty to no better than ± 5 percent so that pursuit of a better value for the constant C is probably not warranted for purposes of measurement.

The experimental work from which the determination of a' was made utilized mixtures of NO in N_2 . Thus, the foreign broadening gas is N_2 . The question arises as to whether other gases which might

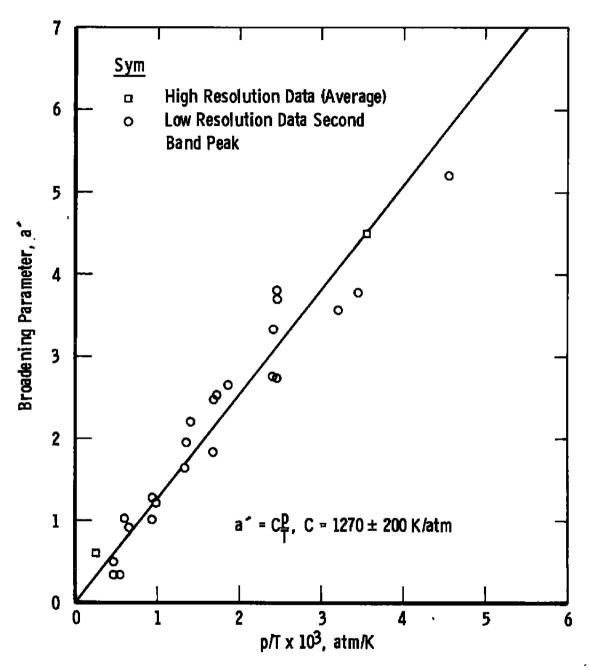


Figure 13. Values of the spectral broadening parameter (a') as a function of p/T for the (0,0) γ -band of NO as obtained by comparing experimental data with computed data.

be found in combustion gas streams, such as CO_2 , H_2O , CO, etc., might have significantly different broadening cross sections. Based on the results presented in Ref. 5 for CO, the broadening cross section for NO is not believed to be significantly different for other molecules than for N_2 . Experience with measurements in absorption cells located in sample lines in which measurements of NO concentration made by other means agreed well with the values obtained by the absorption technique (Ref. 12) also gives confidence to the universality of the broadening parameter with different molecules.

The indirect method of determining the broadening parameter used in this study is believed to be a valid alternative to the use of high resolution spectroscopy to measure line shapes and thus determine the parameter directly. However, the method requires the use of a good model for the radiation source and for the absorbing media, and a high speed digital computer to accomplish the complex numerical calculations.

Finally, a significant result of the work reported herein is the improved value of the effective cross section (or diameter) for collisional broadening of the ground state energy levels of NO. The value of the effective broadening collision diameter (3.5 \pm 0.3 Å) determined in this work offers considerable improvement in uncertainty over the best known previous estimate of Thorsen and Badger (Ref. 2) of 3.8 \pm 1 Å.

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APPENDIX A COMPUTER PROGRAM FOR THE CALCULATION AND SIMULATION OF THE NO (0,0) γ -BAND SPECTRA TRANSMITTED THROUGH AN ABSORBING MEDIUM

The computer program included in this appendix was developed to calculate the transmission of a band of spectral lines from a particular molecule through an absorbing medium containing the same molecular species. The approach may be used for any molecule, but the line structure and angular momentum coupling are so different for each species that a separate program is usually needed for each molecular species. In this appendix, the computational procedure is described for the NO molecule.

DESCRIPTION

The program is separated into two steps. Step 1 records all input data and computes the transmission of the radiation due to each source line. Step 2 processes these data and generates the simulated spectrometer output signal. The program is currently being run on the IBM 370-165 computer; X-Y plotter data are presented on the CALCMP 765 plotter.

PROGRAM STEP 1

The first program step is divided into four parts:

Part 1 — Input Data

Part 1 contains the various COMMON and DIMENSION statements for the program. If more than 500 emission lines are to be considered, a modification of the program would be necessary. Values of $\Delta\lambda$, a', T_a , T_s , and ℓ and various plot control parameters are input. Details of the input data and format are discussed in the following paragraph.

Part 2 — Spectral Line Calculation

Cards identifying each emission-line to be considered are read. For each line, the rotational energy of the upper electron state F(J') is computed via SUBROUTINE FUPPER, and the rotational energy of

the lower rotation state F(J'') is computed in SUBROUTINE FLOWER. The Hönl-London factor is then computed in SUBROUTINE HONNUM. The value of $I_{\nu_j^0}$ is found by interpolation from Fig. 7. The curve in Fig. 7 is an input to the program.

For the (0, 0) γ -Band of NO, the values of k_{ν_i} are given by Eq. (20) and are computed in Part 2.

Part 3 — Line-By-Line Transmission Calculation

In Part 3, the transmission (Eq. (13)) is evaluated for each line. Since k_{ν} is a function of ν , many time-consuming evaluations of k_{ν} are apparent. In fact, if there are n nodes used in the numerical integration, then nm² evaluations are required where m is the number of lines being considered. If the program variable ALLOW is set to zero, the program will sum over all the lines, i.e.,

$$\sum_{i} k_{\nu_{i}} = \sum_{i=1}^{m} k_{\nu_{i}}$$
 (A-1)

To save time (although at the expense of some accuracy), another option is available. This option is selected by setting ALLOW > 0. By considering the jth spectral line, the sum $\sum_{i=1}^{L} k_{\nu_i}$ is written as:

$$\sum_{i} k_{\nu_{i}} \approx \sum_{i=j}^{j+j_{u}} k_{\nu_{i}} + \sum_{i=j-1}^{j-j} k_{\nu_{i}}$$
 (A-2)

Note that the index in the second sum decreases. The indices j_u and j_e are chosen so that the contributions of the $(j + j_u)^{th}$ and $(j - j_e)^{th}$ lines are as small as desired. The exact criterion used is

$$\begin{vmatrix} \frac{k_{\nu_{j+j_u}}}{\sum\limits_{i=j}^{j+j_u} k_{\nu_i}} \end{vmatrix} < ALLOW$$
 (A-3)

and

$$\begin{vmatrix} k_{\nu_{j\rightarrow j_u}} \\ \frac{\sum_{i=j-1} k_{\nu_i}}{\sum_{i}} \end{vmatrix} < ALLOW$$
(A-4)

In effect, only those lines in a chosen neighborhood of ν_j^{\bullet} are considered.

If a' = 0, no collision broadening is considered and k_{ν_i} is calculated from Eq. (10). If a' > 0, then k_{ν_i} must be given by Eq. (A-4).

Subroutine WFUNC evaluates the function

R{exp
$$[-(\omega_j + ia')^2]$$
 erfc $[-i\omega_j + a']$

which is contained in Eq. (9) and leads to the determination of k_{ν_j} . The subsequent evaluation of the integral in Eq. (13) is done numerically using the trapezoidal rule.

Part 4 — Data Storage

This part writes program control data as well as wavenumber and source line transmission data onto a disc file for passage to Step 2.

PROGRAM STEP 2

Step 2 performs the plotting functions of the program. The mathematical and physical considerations are presented in detail in Ref. 1.

The program first reads in the various control parameters from Step 1. Then, the first set of values of ν_j and \overline{T}_j are read in. These data correspond to the first value of N_0 supplied to Step 1. All following curves will be plotted according to the scale factor determined from the initial case. Generally, the case N_0 = 0 (no absorption) is calculated first, and all other plots are referenced to this case (Fig. 1).

If IPLOT1 = 1, then zero slit width plots are produced (Fig. 2).

If IPLOT2 = 1, a separate plot is produced for each value of $N_{\rm O}$ as well as the final combined plot.

The conglomerate spectral profile is constructed as follows: A line with a height proportional to the spectral intensity is drawn at $\lambda_1 + 1/2 \Delta \lambda_X'$ for each spectral line and a triangular slit function of base width $2\Delta \lambda_X'$ is constructed about that line. To arrive at the conglomerate profile, the contributions from each line at a given value of λ are simply added.

PROGRAM VARIABLE DESCRIPTION

The following variables are used in the program:

Mathematical Symbol	Program Variable	<u> Usage</u>
$\overline{\mathtt{T}}_{\mathtt{j}}$	TJ(J)	Transmission of spectral line, j
$ u_{\mathbf{j}}^{\mathbf{\circ}}$	WO(J)	Center wavenumber of jth spectral line
$(\Delta_s \nu_j)_D$	DWJ	Doppler width at half maximum intensity of jth spectral line
(∆a ^ν j)D	DWL	Doppler width at half maximum absorption coefficient $k_{\nu_{j}^{o}}$ of the absorption line
$^{\rm I}_{\nu_{\dot{\rm j}}^{\bullet}}$	E(J)	Intensity of source spectral line
™ j	EO(J)	Intensity of source spectral line at center wavenumber
$^{\mathbf{k}_{\mathcal{V}}}$ i	CAY(I)	Absorption coefficient for the ith line
L	EL	Absorption path length
a*	AP	Collisional broadening parameter
T_s	TE	Source gas temperature, K
$\mathtt{T_a}$	TA	Absorber gas temperature, K
Δλ	SLIT	Equivalent slit width
N _o	ENO	Total number density
v′	IV U	Upper vibrational state
v "	IV L	Lower vibrational state
J "	RJPP	Lower rotational state

Mathematical Symbol	Program Variable	Usage
	W(I)	Nodes for numerical integration (Eq. (1))
	NUP	Upper spin state
	NLO	Lower spin state
	BRANCH	Line branch designation
	IPLOT 1	For zero slit width plots, set IPLOT1 = 1 otherwise = 0
	IPLOT 2	For separate N_0 plots, set IPLOT2 = 1, otherwise = 0
	YHGT	Maximum height of spectral plots, in.
·	DELPLT	Scale for abscissa of spectral plots, A°/in.
8 _J ″ _J ′	S(J)	Hönl-London factor for jth line
F(J')	FU	Rotational energy of the upper electron state
F (J")	FL(J)	Rotational energy of the lower rotational state
	ALLOW	Relative error (see Eq. (10))
m	NLINES	Number of emission lines

DATA INPUT

All data is input to Step 1.

CARDS 1

FORMAT (2F10.0)

Column 1 F(J)

Column 11 $(I_{\nu_i^o}/\delta_{J''J'})$

The first cards expected by the program are values of $(I_{\nu_j^e}/\delta_{J''J'})$ versus F_{u_j} , one data pair per card. The cards should be arranged in order of increasing F_{u_j} . A blank card must follow the final data card of this group. The data for the source lamp used in the measurements shown in this report are given in Table A-1.

CARD 2

TITLE CARD

FORMAT (20A4)

This card should contain any title information the user wishes to use for plot identification.

	CARD 3	DELPLT	FORMAT (F10.0)
	CARD 4	SLIT	FORMAT (F10.0)
	CARD 5	AP	FORMAT (F10.0)
	CARD 6	TA	FORMAT (F10.0)
	CARD 7	TE	FORMAT (F10.0)
	CARD 8	EL	FORMAT (F10.0)
	CARD 9	YHGT	FORMAT (F10.0)
	CARD 10	IPLO T l	FORMAT((I1)
	CARD 11	IPLOT2	FORMAT (II)
بد	CARDS 12	FORMAT (3X, A1, 2I1, 1X,	F4.0, 1X, F11.0)

These cards are those which describe the spectral lines to be considered.

COLUMN 4	BRANCH	(P, Q, or R)	FORMAT A1
,COLUMN 5	NUP		FORMAT I1
COLUMN 6	NUPP		FORMAT II
COLUMN 8-11	RJPP		F4.0
COLUMN 13-23	WO		F11.0

These cards must be arranged in order of increasing wavenumber WO. A blank card must follow the last line description card. The data for the NO (0,0) γ -Band are given in Table A-2.

CARDS 13 FORMAT (F10.0)

These cards contain the values of ENO to be considered. All plots will be scaled to the plot representing the first value of ENO in this group.

SAMPLE CASE

The case to be considered here is the (0,0) NO γ -band with collisional broadening considered. The following values will be used:

$$a' = 1.5$$
 $T_a = 420 \text{ K}$
 $T_e = 320 \text{ K}$
 $\ell = 91 \text{ cm}$
 $\Delta \lambda = 1.6 \text{ Å}$
 $N_o = 0., 1 \times 10^{15}, 1 \times 10^{16} \text{ cm}^{-3}$

YHGT = 10 in.
DELPLT = 6 Å/in.

The equations and constants used for computing the upper and lower energy states and the Hönl-London factors are discussed in Ref. 3 and are repeated here for completeness.

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For the upper state (SUBROUTINE FUPPER),

$$F' = T'_{n} + G' + F'_{n}$$
 $n = 1,2$

where

$$T_{\star}' = 43965.7 \text{ cm}^{-1}$$

$$G' = \omega_e'(v' + 1/2) - \omega_e x_e'(v' + 1/2)^2$$

where

$$\omega_e' = 2374.8 \text{ cm}^{-1}$$

 $\omega_e x_e' = 16.46 \text{ cm}^{-1}$

$$F_{1}' = B_{v}'(J' + 1/2)(J' - 1/2) + D_{v}'(J' + 1/2)^{2}(J' - 1/2)^{2}$$

$$F_{2}' = B_{v}'(J' + 1/2)(J' + 3/2) + D_{v}'(J' - 1/2)^{2}(J' - 3/2)^{2}$$

$$B_{v}' = B_{e}' - \alpha_{e}'(v' + 1/2)$$

where

$$B'_{e} = 1.9977 \text{ cm}^{-1}$$
 $\alpha'_{e} = 0.0198 \text{ cm}^{-1}$
 $D'_{v} = -6.2 \times 10^{-6} \text{ cm}^{-1}$

For the lower state (SUBROUTINE FLOWER),

$$F'' = T''_e + G'' + F''_n$$
 $n = 1, 2$ $T''_e = G'' = 0$

$$F_1'' = B_v''[(J'' + 1/2)^2 - 1 - u] + D_v''J''^4$$

$$F_2'' = B_v''[(J'' + 1/2)^2 - 1 + u] + D_v''(J'' + 1)^4$$

where

$$u = \left[(J'' + 1/2)^2 - Y \sqrt{1 - \frac{Y_v}{4}} \right]^{1/2}$$

 $Y_v = A/B_v^*$

and

$$A = 124.2 \text{ cm}^{-1}$$
 $B_{v}'' = B_{e}'' - \alpha_{e}''(v'' + 1/2)$
 $B_{e}'' = 1.7046 \text{ cm}^{-1}$

$$a_n'' = 0.0178 \text{ cm}^{-1}$$

$$D_{...} = -4.8 \times 10^{-6} \text{ cm}^{-1}$$

The Hönl-London factors are given in Table A-3.

The emission lines being considered can be read from the data card listing which follows. The plots produced by this case are shown in Fig. A-1. Many other calculations as well as comparisons between actual and computed spectra are presented in Ref. 1.

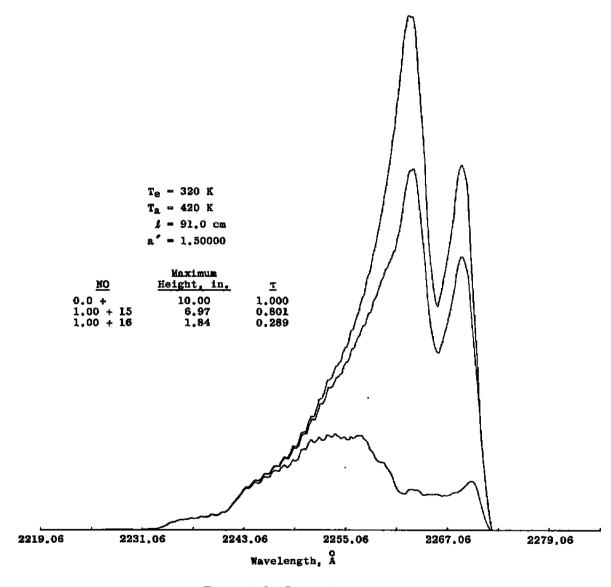


Figure A-1. Spectral test case.

Table A-1. Data Cards for Sample Case

Column	1	48600.	.0218
COLUMN	*	48700.	.0198
		48800.	.018
45100.	1.25	48900.	.0165
45200.	• 92	49000.	.015
45300.	•68	49100.	.0138
45400.	. 54	49200.	.0126
45500.	. 45	49300.	.0115
45600.	• 3 9	49400.	.0105
45700.	.33	49500.	.0097
45800.	• 28	49600.	.0088
45900.	• 255	49700	.0081
46000.	.23	49800	.0074
46100.	·208	49900	.0068
46200.	•19	50000.	.0062
46300.	.175	50100.	•0056
45400.	•158	50200.	.0052
46500.	.145	50300.	.0047
46600.	.1325	50400.	.0043
46700.	.12	50500.	.0039
46800.	-11	50600	.0036
46900.	•1	50700.	.0033
47000.	•092	50800.	.003
47100.	.084	1	2003
47200.	• 077		
47300.	.07		
47400.	• 064		
47500.	•058		
47600.	.053		
47700.	•049		
47800.	.044	1	
47900.	•041		
48000.	.037	•	
48100.	•034		
48200.	.031		
48300.	.0285		
	.026		
48500.	·0238		

Table A-2. Spectra Test Case

Diam's Cond	
Blank Card	R12(1.5) 44086.15
SPECTRA TEST CASE	P22(9.5) 44087.98
6.	Q12(9.5) 44087.98
1.6	Q22(2.5) 44089.56
1.5	R12(2.5) 44089.56
420.	P12(22-5) 44090.28
320.	P22(10.5) 44091.56
91. Branch (Column 4)	Q12(10.5) 44091.56
10. NUP (Column 5)	022(3.5) 44093.39
0 NUPP (Column 6)	R12(3.5) 44093.39
0 RJPP (Column 8-11)	P22(11-5) 44095.70
P12(10.5) 44052.03	Q12(11.5) 44095.70
P12(11.5) 44052.03	
P12(9.5) 44052.03	R22(1.5) 44096.82
ll l	P12(23.5) 44097.18
P12(12.5) 44052.78	022(4.5) 44097.68
P12(8.5) 44052.78	R12(4.5) 44097.68
P12(13.5) 44054.22	P22(12.5) 44100.45
P12(7.5) 44054.22	Q12(12.5) 44100.45
P12(14.5) 44056.16	Q22(5.5) 44102.72
P12(6.5) 44056.49	R12(5.5) 44102.72
P12(15.5) 44058.33	P12(24.5) 44104.59
P12(5.5) 44058.80	P22(13.5) 44105.72
P12(16.5) 44061.30	012(13.5) 44105.72
P12(4.5) 44061.77	R22(2.5) 44105.72
P12(17.5) 44064.66	Q22(6.5) 44108.17
P12(3.5) 44065.40	R12(6.5) 44108-17
P12(18.5) 44068.69	P22(14.5) 44111.50
P12(2.5) 44069.60	Q12(14.5) 44111.50
P12(19.5) 44073.21	P12(25.5) 44112.34
P12(1.5) 44074.26	R22(3.5) 44113.52
P22(3.5) 44077.58	
P22(4.5) 44077.95	Q22(7.5) 44114.15
P22(2.5) 44077.95	R12(7.5) 44114.15
012(4.5) 44077.95	P22(15.5) 44117.84
	Q12(15.5) 44117.84
Q12(2.5) 44077.95	P12(26.5) 44120.71
P12(20.5) 44078.59	Q22(8.5) 44120.71
P22(5.5) 44078.59	R12(8.5) 44120.71
P22(1.5) 44078.59	R22(4.5) 44121.52
Q12(5.5) 44078.59	P22(16.5) 44124.74
012(1.5) 44078.59	Q12(16.5) 44124.74
P22(6.5) 44080.32	Q22(9.5) 44127.68
Q12(6.5) 44080.32	R12(9.5) 44127.68
P22(7.5) 44082.33	P12(27.5) 44129.98
Q12(7.5) 44082.33	R22(5.5) 44130.42
P12(21.5) 44084.33	P22(17.5) 44132.16
P22(8.5) 44084.83	012(17.5) 44132.16
Q12(8.5) 44084.83	022(10.5) 44135.31
022(1.5) 44086.15	
	R12(10.5) 44135.31

Table A-2. Continued

P12(28.5)	
P22(18.5)	
Q12(18.5)	
R22(6.5)	
Q22(11.5)	44143.44
R12(11.5)	44143.44
P22(19.5)	44148.68
012(19.5)	
P12(29.5)	
R22(7.5)	
022(12.5)	
R12(12.5)	
P22(20.5)	
012(20.5)	
P12(30.5)	
R22(8.5)	
022(13.5)	
R12(13.5)	
P22(21.5)	
012(21.5)	
922(14.5)	
R12(14.5)	
R22(9.5)	
P11(9.5)	
P11(8.5)	
P22(22.5)	
012(22.5)	
P11(10.5)	
P11(7.5)	
P11(11.5)	
P11(6.5)	
P11(12.5)	44180.58
P11(5.5)	44180.58
Q22(15.5)	44181.38
R12(15.5)	44181.38
R22(10.5)	44183.02
P11(13.5)	44183.02
P11(4.5)	44183.02
P12(32.5)	44184.08
P11(14.5)	44186.07
P11(3.5)	44186.07
P22(23.5)	44188-15
012(23.5)	44188.16
P11(15.5)	44189.80
P11(2.5)	44189.80
022(16.5)	44192.18
R12(16.5)	44192.18
P11(16.5)	44194.29
	

PI1(1.5)	44194.29
	44194.93
P12(33.5)	
011(2.5)	
P21(2.5)	
011(3.5)	
P21(3.5)	
Q11(1.5)	
P21(1.5)	44197.98
P22(24.5)	44199.21
012(24.5)	44199.21
P11(17.5)	44199.21
011(4.5)	44199.21
P21(4.5)	44199.21
•	44199.21
P21(0.5)	44199.21
011(5.5)	
P21(5.5)	
011(_6.5)	
	44202.36
022(17.5)	
R12(17.5)	
	44203,49
Q21(0.5)	·
P11(18.5)	
Q11(7.5)	
P21(7.5) R11(1.5)	
R11(1.5) Q21(1.5)	
R22(12.5)	
Q11(8.5)	
P21(8.5)	
RI1(2.5)	
	44209.69
R21(0.5)	
P22(25.5)	
012(25.5)	44211.21
P11(19.5)	44211.21
011(9.5)	44212.25
P21(9.5)	44212.25
R11(3.5)	44213.81
Q21(3.5)	44213.81
022(18.5)	44215.46
R12(18.5)	44215.46
011(10.5)	44217.00
P21(10.5)	44217.00
R21(1.5)	44217.60
P11(20.5)	44217.73

Table A-2. Continued

	Iable 4-4.	Continueu	
R11(4.5)	44218.55	R11(10.5)	44260.93
Q21(4.5)	44218.55	Q21(10.5)	44260.93
R22(13.5)	44220.78	R21(6.5)	44261.79
Q11(11.5)	44222.24	P22(29.5)	44254.24
P21(11.5)	44222.24	012(29.5)	44264.24
P22{26.5}	44223.95	R22(16.5)	44264.24
012(26.5)	44223.95	Q11(17.5)	44266.44
RII(5.5)	44223.95	P21(17.5)	44266.44
021(5.5)	44223.95	Q22(22.5)	44268.65
R21(2.5)	44225.14	R12(22.5)	44268.65
P11(21.5)	44225.26	R11(11.5)	44269.92
022(19.5)	44228.00	021(11.5)	44269.92
R12(19.5)	44228.00	P11(26.5)	44271.53
	44228.00	R21(7.5)	
P21(12.5)	44228.00	011(18.5)	
	44230.23	P21(18.5)	= -
Q21(6.5)	44230.23	P22(30.5)	_
	44233.24	Q12(30.5)	
	44233.31	R22(17.5)	
	44234.47	R11(12.5)	
	44234.47	021(12.5)	44279.65
P21(13.5)	44234.47	P11(27.5)	=
	44236.86	922(23.5)	-
_ : - : : :	44236.86	R12(23.5)	
R11(7.5)	44236.86	R21(8.5)	
021(7.5)	44236.86	Q11(19.5)	
Q22(20.5)	44241.13	P21(19.5)	
R12(20.5)	44241.13	R11(13.5)	44290.12
P11(23.5)	44241.68	921(13.5)	44290.12
011(14.5)	44241.58	P22(31.5)	
P21(14.5)	44241.68	012(31.5)	44294.38
R21 (4.5)	44242.11	R22(18.5)	44294.38
R11(8.5)	44244.30	P11(28.5)	44294.38
021(8.5)	44244.30	R21(9.5)	44295.79
R22(15.5)	44249.10	Q11(20.5)	44296-87
Q11(15.5)	44249.10	P21(20.5)	44296.87
P21(15.5)	44249.10	Q22(24.5)	44298.56
P22(28.5)	44250.17		
012(28.5)	44250.17	R12(24.5) R11(14.5)	44298.56
P11(24.5)	44251.31	Q21(14.5)	44301.17
R11(9.5)	44252.05	P11(29.5)	44306.57
021(9.5)	44252.05	Q11(21.5)	
R21(5.5)	44252.05	P21(21.5)	44308.22
022(21.5)	44254.52	R21(10.5)	44308.22
R12(21.5)	44254.52	P22(32.5)	44308.22 44309.89
Q11(16.5)	44257.54	Q12(32.5)	
P21(16.5)	44257.54	R22(19.5)	44309 ₄ 89
P11(25.5)	44260.93		
11/5313)	77600933	R11(15.5)	44312.82

Table A-2. Continued

	44710 00		
Q21(15.5)		021(20.5)	
022(25.5)	44314.35	R21(15.5)	
R12(25.5)	44314.35	R22(23.5)	
P11(30.5)	44319.69	Q22(29.5)	44382.93
011(22.5)	44320.28	R12(29.5)	44382.93
P21(22.5)	44320.2A	011(27.5)	44389.23
R21(11.5)	44321.38	P21(27.5)	44389.23
R11(16.5)	44325.12	P11(35.5)	44393.03
021(16.5)	44325.12	R11(21.5)	44395.56
P22(33.5)	44326.20	021(21.5)	44395.56
012(33.5)	44326.20	R21(16.5)	44396.60
R22(20.5)	44328.22	P22(37.5)	44397.04
Q22(26.5)	44330.62	012(37.5)	44397.04
R12(26.5)	44330.62	Q22(24.5)	44401.45
P11(31.5)	44332.92	R12(24.5)	44401.45
011(23.5)	44332.92	R22(24.5)	44401.45
P21(23.5)	44332.92	011(28.5)	44404.77
R21(12.5)	44335.22	P21(28.5)	44404.77
R11(17.5)	44337.97	P11(36.5)	44409.50
Q21(17.5)	44337.97	R11(22.5)	44411.51
P22(34.5)	44343.01	021(22.5)	44411.51
Q12(34.5)	44343.01	R21(17.5)	44413.05
R22(21.5)	44345.86	P22(38.5)	44416.27
Q11(24.5)	44345.86	Q12(38.5)	44416.27
P21(24.5)	44345 • 85	022(31.5)	44420.73
022(27.5)	44347.45	R12(31.5)	44420.73
R12(27.5)	44347.45	R22(25.5)	44421.02
P11(32.5)	44347.45	011(29.5)	44421.02
R21(13.5)	44349.51	P21(29.5)	44421.02
R11(18.5)	44351.46	P11(37.5)	44426.61
Q21(18.5)	44351.46	R11(23.5)	44428.01
011(25.5)	44359.81	021(23.5)	44428.01
P21(25.5)	44359.81	R21(18.5)	44430.56
P22(35.5)	44360.38	P22(39.5)	444 35 . 89
012(35.5)	44360.38	012(39.5)	· · · ·
P11(33.5)	44361.81	011(30.5)	
R22(22.5)	44363.61	P21(30.5)	44437.77
022(28.5)	44365.00	022(32.5)	44440.23
R12(28.5)	44365.00	R12(32.5)	44440.23
R21(14.5)	44365.00	R22(26+5)	44441.26
R[1(19.5)	44365.55	P11{38.5}	44444.40
021(19.5)	44365.55	R11(24.5)	44445.05
Q11(26.5)	44374.12	Q21(24·5)	44445.05
P21(26.5)	44374.12	R21(19.5)	44448.65
P11(34.5)	44377.09	011(31.5)	44455.09
P22(36.5)	44378.49	P21(31.5)	44455.09
012(36.5)	44378.49	Q22(33.5)	44460.45
R11(20.5)	44380.21	R12(33.5)	44460.45

Table A-2. Concluded

TUNO IN E	
R22(27.5) 44462.07	Q21(31.5) 44581.73
P11(39.5) 44462.71	R21(26.5) 44591.53
R11(25.5) 44462.71	022(39.5) 44593.46
Q21(25.5) 44462.71	R12(39.5) 44593.46
R21(20.5) 44467.16	011(38.5) 44593.46
Q11(32.5) 44473.15	P21(38.5) 44593.46
P21(32.5) 44473.15	R22(33.5) 44598.17
Q22(34.5) 44481.26	R11(32.5) 44603.58
R12(34.5) 44481.26	Q21(32.5) 44603.58
R11(26.5) 44481.26	R21(27.5) 44614.41
Q21(26.5) 44481.26	Q11(39.5) 44615.72
R22(28.5) 44483.40	P21(39.5) 44615.72
R21(21.5) 44486.41	R22(34.5) 44622.97
	R11(33.5) 44626.16
Q11(33.5) 44491.75	Q21(33.5) 44626.16
P21(33.5) 44491.75	R21(28.5) 44637.85
R11(27.5) 44500.09	R22(35.5) 44648.22
Q21(27.5)_44500.09	R11(34.5) 44649.12
Q22(35.5) 44502.61	Q21(34.5) 44649.12
R12(35.5) 44502.61	R21(29.5) 44661.86
R22(29.5) 44505.24	R11(35.5) 44672.76
R21(22.5) 44506.13	021(35.5) 44672.76
Q11(34.5) 44510.92	R22(36.5) 44673.94
P21(34.5) 44510.92	R21(30.5) 44686.59
R11(28.5) 44519.47	lt.
Q21(28.5) 44519.47	R11(36-5) 44697-12
Q22(36.5) 44524.55	Q21(36.5) 44697.12
R12(36.5) 44524.55	R22(37-5) 44700-02
R21(23.5) 44526.75	R21(3145) 44711.69
R22(30.5) 44527.76	R11(37.5)_44722.14
Q11(35.5) 44530.67	Q21(37.5) 44722.14
P21(35.5) 44530.67	R22(38.5) 44727.04
R11(29.5) 44539.74	R21(32.5) 44737.46
Q21(29.5) 44539.74	R11(38+5) 44747+37
Q22(37.5) 44546.76	Q21(38.5) 44747.37
R12(37.5) 44546.76	R22(39.5) 44754.43
R21(24.5) 44547.80	R21(33.5) 44763.28
R22(31.5) 44550.89	R11(39.5) 44773.43
011(36.5) 44550.89	Q21(39.5) 44773.43
P21(36.5) 44550.89	R21(34.5) 44789.81
R11(30.5) 44560.38	R21(35.5) 44818.26
Q21(30.5) 44560.38	R21(37.5) 44874.81
Q22(38.5) 44569.56	R21(38.5) 44904.54
R12(38.5) 44569.86	R21(39.5) 44934.51
R21(25.5) 44569.86	Blank Card
Q11(37.5) 44571.93	
P21(37.5) 44571.93	0.0
R22(32.5) 44574.03	.000E15
R11(31.5) 44581.73	+000E16

TABLE A-3 HÖNL-LONDON FACTORS FOR $^2\Sigma \rightarrow ^2\pi$ TRANSITIONS INTERMEDIATE BETWEEN HUND'S CASES (a) AND (b)

$$\begin{split} R_{22} &= \frac{(2J''+1)^2 + (2J''+1)[\gamma(\gamma-4) - (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8(J''+1)} \\ Q_{22} &= \frac{(2J''+1)[(4J''^2 + 4J''-1) + [\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(8J''^3 + 12J''^2 - 2J''+1-2\gamma)]}{8J''(J''+1)} \\ P_{22} &= \frac{(2J''+1)^2 + (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''-7+2\gamma)}{8J''} \\ R_{12} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''-7+2\gamma)}{8(J''+1)} \\ Q_{12} &= \frac{(2J''+1)[(4J''^2 + 4J''-1) - [\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(8J''^3 + 12J''^2 - 2J''-7+2\gamma)]}{8J''(J''+1)} \\ R_{11} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ Q_{11} &= \frac{(2J''+1)^2 + (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 + (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''+1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''-1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''-1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''-1-2\gamma)}{8J''(J''+1)} \\ R_{21} &= \frac{(2J''+1)^2 - (2J''+1)[\gamma(\gamma-4) + (2J''+1)^2]^{-\frac{1}{2}}(4J''^2 + 4J''-1-2\gamma)}{8J''(J$$

where $Y = A/B_u$

THE PROGRAM

```
IV 6 LEVEL 21
                                                                                                                      DATE = 75178
                                                                                                                                                                        09/06/45
                                                                         MAEN
         т.
           C PARTI PARTI PARTI PARTI PARTI PARTI PARTI PARTI CERRORIO CONTROL CON
                          REAL#4 KO
                           INTEGER+4 P.G.R. ELANK
                          INTEGER+4 BRANCH
DIMENSION S(500) FL(500)+WC(500)+E0(500)
                         DIMENSION HEADS(20)
                          DIMENSION KO(500)
                          DATA P/'P'/.Q/'G'/.A/'R'/
                          DATA BLANK/ 1/
                          COMMON/PAREM/CONSTI, CONSTI, CENSTI, WOLEO. KO, NLIMES
                          DIMENSION W( 500) . E( 500) . CAY( 500) . DAT ( 500) . EJ( 500) . TJ( 500)
                       CALL ERRSET(208-256-1.1)
                          ALLOW- . GOL
                       CALL FUNCEXDUM
                          L 182=2
                    ___I_VU= 0
                          IVL=0
                          READ(5.366)HEADS
                366 FORHAT (20A4)
                     __READ( 5.36)0ELPLT
                          READ(5,36)SLIT
                          ULAM=SLIT
                       READ(5,36)AP
                        READ(5,36)TA
                          READ(5.36)TE
                       READ($,36)EL
                          READ(5.36) YHGT
                   READ(5.700)1PLOT1
                          READ(5.700)[PL012
                700 FORMAT(211)
                    37 FURMAT(11.2F10.0)
                          C1=-601550-14/TA++(3-D0/2-D0)
                           C L=C 1+2.00
                          C2-1.4383600/TA
                          CUNSTI=-2.DO+DSQRT(DLGG(2.DQ))
                          CONST2=1.3670-74 SQRT(TE)
                           CONST3=1.3070-74 SQRT(TA)
                           CONSTA=J.E-7+SORT(TE)
                           WRITE(6.87]HEADS,AP
                   [4.73.=*AHE.KD1, 4ADS.KOS.IHI}TAMRD7 TB
           C+++
            C PART2             C+++++++++++++++++++++++++++++++++++
                                                                                                       *****************************
            C
                 749 READ(5.1)BRANCH, NUPINLE . RJPP. WC(J)
                          IF (BRANCH-EG-BLANK)GO TO 2
                           JPP=HJPP
                      1 FGRMAT(JX.AL.11.11.1%.F4.0.1%.F11.0)
                          'IF(BRANCH.EQ.P)GO TO 16
                           IF(URANCH-EQ-Q)GO TC 2C
```

```
IN & PEART ST
                            MAIN
                                            DATE = 75178
                                                                09/06/45
         CALL FUPPER(IVU-NUP-JPP+1-FU)
        CALL HONNUME I VU. I VL , NUP . NLO . JPP+ 1 . JPP .3 . S(J) }
          GD TO 100
       JO PURMAT(AFIG.0)
       20 CALL PUPPER(IVU-NUP-JPP -FU)
          CALL HOMNUMCIVU.[VL. KUP. KLC. JPF. JPP. 2.5(J))
          GU TO 160
       10 CALL FUPPER(IVU, NUP. JPP-1.FU)
         CALL HOMNUM(IVU.IVL.ALP.NCC.JPP-1.JPP.1.S(J))
     100 CALL FLOWERITYLINGS, JPP.FL(J)]
          EO(J)=S(J)*FUNC(FU)
          TRITE(6,101)J. WRANCH, NUP, NLC, JPP, FU, FL(J), WU(J), S(J), EU(J)
      101 FURNAY(2x,13,2x,11,11,11,4( ",12, ") ", "FU= ",016,8,5x, "FL=",D16,8;
         +5x.'#D=',D16.8,Ex,'SJ=',D10.8,5x,'EQJ=',D16.8)
         J÷J+L
          GO TO 749
        2 CONTINUE
         NL INESEJE I
      800 READ(5.30,END=800)ENC
         DU 69 I=1.NLINES
         B=C1+S(1)+ EMP(C2+FL(1))
         KO(1)=8*ENU
       69 CUNTINUE
    Ċ
          ISAVE= 1
         NINTEAC
          NP1=NLNT+1
          FAUTOP=0.00
          TAGEDT=0.DQ
         DU 200 J=1.NLINES
         DaJ=CONST2+WU(J)
         DELTAY=CONST4##04J3
          AZJENU(J)-DELTAN
         B∠J= eO(J}+DELTA m
         DIST=bZJ-AZJ
         DEL=DIST/NINT
         DU 7646 I=1.NP1
         W(1)=AZJ+(I-1)+DEL
         E(1)=E0(J)+ EXP(-((W(1)-WC(J))+CCNST1/DWJ)++2)
          1F(ENG.EQ.0.3GO TO 67447
         JLEFT=0
         SLML=0-
         SUMR-0.
         [[±]
     1108 LWL=CGNST3+BO(111)
          IF(AP-EU-0-)GD TO LICC
         deta=-consti#(#(1)-#C(11))/D#L
         LALL BEUNCIGETA . AP . TERM . DURS)
         Cu Tu 1101
     1100 TERM=EXP(-{( m( I )-4G( I I ) )*CONST1/DeL) **2)
     1101-CONTINUE
         THEW-KCCII JOTLAN
```

```
09/06/45
IV G LEVEL 21
                            MAIN
                                             DATE = 75178
          lfijleft.Eq.1)GG to 11GS
          SUMM-SUMR+THEW
          IFEWO(11).GT.W(1).AND.TNEW.EG.O.)GO FC 1107
          1F(11.EQ.MLINES)60 TO 1107
          1f(wu(11).LT.BZJ)GC TO 33550
          IF(AdS(THER/SUMR).LE.ALLCH)GC TO 1107
    33550 CUNTINCE
          11=11+1
          GU TO 11 CE
     1107 JLEF (= L
          1 (=..-1
          EIDS OF 00(1-04-1)41
          GU TU LLON
     1109 CUNTINUE
          SUML=SUML+INEN
          [F(#O(11)-LT.#(1)-AND.TAEB-EQ-U-JGC TO 7013
          IF(11.EQ.1)GG TU 7413
          IFIBU(II).GT.AZJIGO TC 33551
          IF(AUS(THEM/SUML)-LL-ALLCH)GO TO 7013
    33551 CONTINUE
          14=11-1
          GL TO 1108
     7013 CONTINUE
          60 TO 7000
    67447 SUML=L.
          SUMMEO.
     TGUU CAY( I )= SUML+SUMR
          SUM1=6.00
          5UM2= 0.00
          ひとと とこいをし / と。ひ 0
          00 9006 1=1.NP1
     YOUR DAT(1)=E(1)+ EXP(-CAY(1)+EL)
          THIM. SEL SOUP UC
          SUM 1= SUM 1+C=L2+(DAT( 1)+DAT( 1+1) 1
     9001 SGM2=SGM2+ULL2*(E(1)+E(1+1))
          と」と」 P=SUM2
          TJ(J)=SUM1
          fALTUP=IAUTOP+TJ[J]
          [L]L34TuduAT=TUBUAT
      200 CONTINUE
    C PART4 PART4 PART4 PART4 PART4 PART4
    C
          TALETALTUP/TAUSUT
          ALPHA= L.UU-TAU
          WRITE(L. 848) TA. FL. ENG. CL. TAL. ALFHA
      EGH FLAMAT(//, ta. 'TA=" .Diu-cada, "Te=".Diu-cada) "NC=".Diu-cada) "L=".
         FUIC. E. DA. / . DA. ! TAUE! . DIG. B. DX. !ALPHA=!. DIG. B.
          ARTITE( 20) SELT, PETPIN . FETPAR, DEEPLT . DEAM . YMGT . ALIRES , END.
         *TL.TA.EL.IFLUTI, IPLLT2
         * . L ARZ . AP
         . IAL
         4. FLAUS
          UN 5431 [=1.NLINES
```

IV & LEVEL 21 MAIN DATE = 75178 09/00/45

LKI=NLINES-I+1

9431 AKITE(20]NO(LKI),TJ(LKI)
GU TO 866

600 ENDFILE 20
REWIND 20
STOP
END

IV G	LEYEL	21	FUNC	DATE = 75178	09/06/45
		FUNCTION	FUNC(X)	· — — — — — — — — — — — — — — — — — — —	
		DIMENSIO	N XDATA(18C).YDATA(100)		
		DATA IRS	T/0/		
		IF (IRST .	EQ-01GO TO 1		
		IF(X.LT.	XDATA(11)GO TO 200		
		DO 10 1=		<u> </u>	
		IF (XDATA	[] + LE - X-AND - XDATA[] + [] -	GE.XIGO TO 20	
	10	CUNTINUE		•	
		FUNC=(YD	ATALNPI-YDATÁLNPL)) •(X-X	DATA (NP\$) / (XDATA (NP) - XI	Y+ ((CAN) A TAC
	_ '	·UAȚA(NP.)			
		RETURN			
	20	FUNC= (YD	TADX) \ ({ 1 } ATAGY — (1 + 1) ATA	MX-X3 = ({ 1 } A TADX - (1 + 1) A	CC13ATA
	1	1)ATAGY++)		
		RETURN			
	200	FUNC=(YD	ATA(2)-YDATA(1)}+(X-XDAT	A(1) 1/(x DATA(2)-XDATA(1) 1+YDATA(L)
		RETURN			
	1	Cu 100 I	=1-1000		
		HEAD(5.2	[1]ATAQY.[1]ATAQKE		
	2	FORMAT(8			
		[F(XDATA	(1).EQ. 0.DC)GQ TQ 3		
		CUNTINUE			
	.3	CUNTINUE			
		NP= I - 1			
		Nb T=Nb→ t			
		IRST=1			
		KETURN			
		END			

```
IV G LEVEL 21
                                FUPPER
                                                  DATE = 75178
                                                                        09/06/45
           SUBROUTINE FUPPER(V.M.JINDEX.F)
    ٠<u>د</u> .
           SUBMOUTINE FUPPER CALCULATES THE ROTATIONAL ENERGY OF THE UPPER
           ELECTRON STATE FOR A GIVEN VIBRATIONAL STATE(V). A GIVEN SPIN STATE(M). AND A GIVEN ROTATICAAL STATE(J)
    C
    C
           INTEGER V
           REAL#4 J
           J=J INDEX+. &
           T*43965.7
           WE=2J74.8
           ₩E XE= 16.46
           BE=1.5977
           ALPHAE=.0158
           UV--6.2E-6
          BY=BE-ALPHAE+( V+.5)
           GD TO(1.2).M
         1 FN=BY+(J+.5)+(J-.5)+DY+(J+.5)++2+(J-.5)++2
           GU TO 3
         $ FN=8v=(J+.5)+(J+1.5)+DV+(J+.5)++2+(J+1.5)++2
         3 CONTINUE
           G=#E+(V++5 |- WEXE+(V++5) ++2
           F=I+G+FN
           RETURN
           END
```

IV G LEVEL ZI FLUNEH DATE = 75178 09/06/45 SUBROUTINE FLOWERLY . M. JINDEX . F) SUBROUTINE FLUWER CALCULATES THE ROTATIONAL ENERGY OF A GIVEN RUTATIONAL STATE(J). YIBRATIONAL STATE(Y). AND SPIN STATE(M) c INTEGER V REAL#4 J J=JINDEX+.5 T=0. G=0. UY=- 4 . EE- 6 A= L<4.2 HE=1.7046 ALPHAL=.0176 EV=BE-ALPHAE+(V+.5) VB\A=VY U=SURT((J+.5)+42-YV+([.-YV/4.)) GD TU(1.2).M L FN=uV+((J+.5)++2-1.-U)+DV+J+44 GO TO 3 2 FN=644((J+.5)*#2-1+U)+D4#(J+1.)*## #=1+G+FN RETURN **ENO**

```
IV G LEVEL 21
                               HENNUN
                                                 DATE - 75176
                                                                       09/06/45
           SUBROUTINE HONNIM(NUP.NLPP.MEGAN.MEGAN.JF.JPP.IB.HONL)
           SUBROUTINE HONNUM CALCULATES THE HONL-LOADEN FACTOR WHERE
     c
           NUPSUPPER VIBRATIONAL STATE
          NUPP-LOWER VIBRATIONAL STATE
     ċ
           MEGAN-UPPER SPIN STATE
    c
          MEGAM-LUMEN SPIN STATE
    Ç
          J=100
    ¢
           IN-I FOR P BRANCH. -2 FCR Q BRANCH. -3 FOR A BEANCH
    Ċ
         REAL#4 J
          J=JPP+.5
           T1=2.+J+1.
           T##T1#11
          TJ=1./SURT(73.3+69.3+12)
           JHE [H+1
           GG TU(2.2.3.4),JB
         2 CUNT INUE
           GO TO($1.22).MEGAN
        21 CONTINUE
           GL TD(211.212). MEGAM
       211 RNUM=T2+$14T3+(4.4J4J+4.4J+1.-146.5)
          DENOM=8.#J
           GU TO 11111
                              -- --
       212 RNUM=T2-T1+T3+44.+J+4.+J+1.-L46.5)
          DENUM= 8.4J
           GO TO 11111
        22 CONTINUE
          GU TO( 221, 222) . MEGAM
       221 RNUM=T2-11+T3+(4.+J+J+4.+J-7.+146.5)
          DENOM=8.4J
          GL TO 11111
       222 RNUM=T2+T1+T3+(4.+J+J+4.+J-7+146.5)
          DENDM=8.+J
          --GO- TO -111:1:1-1--
        3 CONTINUE
          GG TO(31.32).MEGAN
        31 CUNTINUE
          GO TO(311.312). MEGAM
       311 RNUN=T1+([4+J+J+4+J-1]+T3+(8.+J+3+12.+J+J-2.+J-7.+146.5))
      DENDM=#. #J#(J+1.)
                                       ----
          GO TO 11111
       312 RNUM=TI+(4.*J+J+4.*J-1.-T3+(8.*J+3+12.*J+42-2.*J-7.+146.5))
          UENOM= 8. # J + [ J + ] . }
          GC TO 11111
        32 CONTINUE
          GO TO(321,322). MEGAM
       321 RNUH=T1+((4.0J0)+4.0J-1.;-130(8.0J0+3+12.0J0)-2.0J-1.-146.51)
          DENOM=8.4J*(J+1.)
          GO TO 11111
      ({d.aaf--1+L*.s-L*L*.s-1+E**L*.s)+E**L*.s-1+L*.s+L*.s-1+1*13=MUNR 22E
          DENOM=8.*J*(J+L.)
          GO TO 11111
       4 CONTINUE
          GO TOL41,42) MEGAN
        41 CUNTINUE
          GO TO(411.412).MEGAM
```

```
IV G LEVEL 21
                           HUNNUP
                                           DATE = 75178
                                                              04/06/45
      411 KNUM=T2+11+T3+(4.+J4J+4.4J-7.+L46.5)
         DENUM=2.#(J+[.)
         GG TU 11111
      412 KNUM=T2-T1+T3+[4.+J+4.+J-7.+[46.5]
         DLNUM= 6. + (J+1)
         Gu Tu 11111
       42 CONTINUE
         GG TJ(421.422).F2GAM
      DENUM=E. +( J+I.)
         36 TU 11111
      422 -1+L4-+4-+1+1-111+1-140-5
         DENUM=6.+(J+1.)
    IIIII CONTINUE
         If (GENGMAEGAG)GC TE SAI
         HUNL=RNUM/UENJM
         RLICKN
      591 HUNL=C.
         RETURN
         END
```

IN G CCAFF	-l	WFUNG	DATE = 70176	U 3/06/45	
	SLOKULTINE	mFUNC(Xf eYf emighe)		w(2)	1
ر ن ن	W L . W 2 ARE T	HE HEAL AND IMAGINAR	Y PANTS OF THE ARGUMENT Y PARTS OF THE FUNCTION NATIONAL BUREAU CF STAND CNS, PAGE 297)	AFOS	
·	CUMPLEX#10 CUMMON/TAUL	AL+d(A-+,L-2) 2C,HC,J,2S,T1,72,T3 L/2(31,4L,2),X(4U),Y	(31)	W(Z) W(Z) W(Z)	2 3 4
		(2) } flag= flag+ flag= flag+3		W{Z} W(Z} h(Z) W(Z)	5 6 7 8
		90v.UR. YI.GE.3.D0)G0	TG 1	W(Z) W(Z) W(Z)	9 10 11
	11=11=111 1+1=111 1=10+x1+1 1=10+x1+1			W(Z) W(Z) W(Z) W(Z)	12 13 14 15
	Z2=(Z(1[1.J	J•K}-2(III,JJ1•K))*(*	rt-×t17f3)\text171} -xt17f3)\text17f2\\ -xt17f3)\text17f2\\	W(2) 2(11-JJ1-K)W(2) W(2)	16 17 18
	# (K) = (Z) - Z 2 # L = # (1) # (K) = (Z) - Z 2 # (Z) 1 h + J J (h)) +(Y]-Y(]]]))/(Y(]])-	-Y(111))+22	W(2) W(2) W(2) W(2)	19 20 21 22
	RETURN	DC.3CC.4QCJ.IFLAG		W(Z) W(Z) W(Z)	23 24 25
. '	J=(0=0 C=1 =0= ZC=DCMPL x(x ZS=2C+2C			W(Z) W(Z) W(Z) W(Z)	26 27 28 29
	TZ=+6499421) C/(251 \$ C1635DQ) LOU/(25-1.7844 \$27DQ) \$4DQ/(25-2.5253437QQ)		W(Z) W(Z) W(Z)	30 31 32
	# L= (WL +JCON, #z=(DCGNJG (! IF (IFLAG .CG.	G(#C)		W(2) W(2) W(2) W(2)	33 34 35 36
2	Gu Tu(100,20 J=(0.00,1.00 ZC=DCMPLx(x ZS=ZC+ZL			n(2) W(2) W(2) W(2)	37 36 39 40
	T2==0:176:34 #L=J*ZC*{T1:)0/(25275255100) 50u/(25-2.72474500) FT2) JG(MC))*.506		W(Z) W(Z) W(Z) W(Z)	41 42 43 44
	m2=(UCDNJG(If(IFLAG.∈G: GC TO(IGC+3c	.C}-#C +J+.≒DQ		W(Z) W(Z) W(Z)	45 46 47
	# 2 = - # 2 # E T UK N G = 2 • D J # D E XP ((1x+1x-1x+1Y		W(Z) W(Z) W(Z)	48 49 50

AEDC-TR-76-12

IA & FFAFF	ا ب	nt UAC	JATE = 75178	04/06/45	
	GG=2.0G*AL	*Y]		w(Z)	51
	#1=6≠ひししらし	G6)- = 1		w{Z}	52
	#Z=G+U5LN(G6]+a2		#(Z}	5.3
	イビョレック			W4.2)	54
4.36	G=2.00+ot X	P(YI+YI-XI+XI)		w(2)	55
	Gu=2.00*x1	+ Y I		w12)	56
	# L=0#0CD5(*	G 6) - w 1		W(Z)	57
	#2=-0+0516	(GG)-#2		W(Z)	56
	KL TUKN			W(Z)	59
	E-4D			a(Z)	60

IN C FEAST SI	BLK DATA	DATE = 75176	09/06/45
BLOCK DATA			W(Z) 61
	AL*8(A-H,L-Z)		W(Z) 62
CLMMON/FAJL		.4213 4	w(Z) 63
			u(31),4 7(31),8(4)
			J(31].Zl4(31),W(Z) 66 O(31).ZZJ(31).W(Z) 66
_			/(J1),225(J1),w(2) 67
			4(31).235(31).W(Z) 66
			1(11),242(31), w(2) 69
			b()1),24y(31),#(2) 7u
			5(J1).25c(J1).w(2) 71
#457(31), ¿			2(31),203(31),a(Z) 72
#264(31). Z	65(J1), 266(Ji), Z6	/lull. 200(31). 20	9(11).270(31).W(Z) 73
#Z71(31), z			o(31).277(31).w(2) 74
*278(31). ¿	79(3[), Zevi3[],		4(Z) 75
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* .04393000; * .07504300;	.05126400.	.05804600,	.36426600,	.06992780,	W(Z) 310
+ .09323900.	- £7963200.	- 083716Du .	.087328DQ,	.09049200,	W(Z) 311
+ .1017e3D0.	. (\$5651DQ.	• 09760d00 •	.09928400,	.100671D0.	W(Z) 312
• .1041,500,	-102649DJ.	-103293DD.	-10373700.	.104002Du,	W(Z) 313
+ .10276700/	-104G66DD.	-10387800.	-10361700.	.1u323600.	W(Z) 314
VOES ATAD					W(Z) 315
+ .0002z3Du.	. CC#769DC.	.017134D0.	.02522500.	.032967D0.	W(Z) 316
• .0403U4DU.	.047194DU.	.05361100.	.05954300.	.052907D01	W(Z) 317 W(Z) 318
* .06954400.	.C74431DG.	.07846200.	.002059D0,	.085245D0.	w(Z) 319
* .0854400.	.09U482DG.	09258400	.U94376D0.	-09588200.	m(Z) 320
+ .05712700.	. 05513360,	.058922DJ.	.09951300.	*099925DO*	■{Z} 320
* .100177DO.	·10028400,	-140261001	.10012200.	-09987900.	w(Z) 322
4 .09954400/			**********	-4,,,,,,,,,,,,	W(Z) 323
DATA Z31/					m(Z) 324
• .udestube. •	.GC794306.	.C1562700.	.02309500.	.u30279D0.	w(Z) 325
# .UJ7126D0.	.U4359866.	.04966500,	.05531100,	.u6052900,	#(Z) 326
* . 46531800,	.Ca9t8500,	.07364100.	.07720200.	.08038500.	W(Z) 327
* .063210Du,	. Cese57DJ.	.06787GDG.	. U89749D0.	.09135500.	#(Z) J28
# .0y2711b0.	. 67363500.	.094748D0.	.04546700.	.096010CO.	W(Z) J29
* .0.039JDJ.	.096632Du.	.69673900.	.09672900.	.09661330,	P(Z) 330
* .0vo462Du/					¥(Z) 331
DATA Z32/					w(Z) 332
.C0000700.	.00725400.	.01433800.	.021250D0.	•04792900•	■(2) 333
• .03415f00.	. L4U4U7DO.	.04614100,	• 051 509D0 •	.05650100.	#(Z) 334
+ .Db]114D0.	.C6535000,	.06921600.	.07272200.	.07588300.	w(Z) 335
+ .U787120U.	. LEL 229DG.	. 08J450D0 ,	-08539400,	.00708000,	w(2) 336
• .00852500.	.CES749DO.	.09076700,	.09159700.	.092255DO,	₩(2) 337
• .19275400.	·C63110D0.	. C93336DO.	.043442D0.	.093442Du.	€(Z) 33 8
• . C93345DQ/					= (2) 339
DATA 233/					#(2) 340

IA C FEAFF	~1	BLK DA	41A	DATE = 75178	09/00	/45
			.01322500.	.01963900.	.02586200,	#(Z) 341
*	.03184900.	.03/565DC.	.04298300.	.04806380.	.05285400.	#(2) 342
	-US72cyuu.	.06118700.	.Qe515100,	.00858700.	.07171100.	m(Z) 343
*	.u7452900.	• 677 CSSDO •	.07930600.	. 181 297 00.	.083044Cu.	#(Z) 344
	•U645¢200₁	.CH5 tu7UJ.	.08047400.	.08790000.	.v88657üu.	#{Z} 345
	* Co 4 % = 400 u	.02571960.	.04005400	********	*040709*00*	w(Z) 346
•	.G90375U0/					w(2) 347
	DATA Z 347					W(Z) 348
			.01225200.	*01855500*	.024032NO.	a(2) 349
	.02904300.	. 03502200.	.C4014400.	.U44989DU.	•u495440U+	W(2) 350
	.05163100.	.05775706.	.06141330.	.06477300.	.067844D0.	w(2) 351
	. 0702 3600.	. 13156000	• v7542300 •	.u/7445DU,	.079236D0.	w(Z) 352 W(Z) 353
	.08081100.	• G#21620G •	. Qd3364DV .	.J84370D0.	.08521JD0.	
	.005.46500.	- U E C 45 E U C +	. 4555534.	.0071y000.	.06739100.	W(Z) 354 W(Z) 355
	.007453DU/					W(Z) 356
	DATA 235/ .cueutodo.	. 60572000.	•011394JJ.	.01696600.	.022403D0.	W(2) 357
	.027c7000	. 61273800.	.03758200	.34218500,	.04653200.	w(Z) 358
	.02767000;	-C54426DC+	.05/971Du .	.00124000,	.06425880.	W(Z) 359
	.00701200	.Co5516DC.	.07178500	.07302300.	+07564600	w(Z) 360
	.07726300.	.C78687D0.	. 67493000	.00100400.	.08192100.	10L (S)W
	.08264600	.08332400.	.00383200.	.064225DO.	.08451100.	#(Z) 362
	.Lc47(CDU/		1000051007	*********	1004011011	W(Z) 363
	DATA Z36/	•				W(Z) 364
	********	.0G5340D0.	.01063300.	.01584600,	.02094400.	m(Z) 365
	.02505700.	. CJUE/7UU.	.03526300	.3963700.	.043785DO.	w(Z) 366
	.34/65200.	.05137CDG.	-054798DU+	.05798400.	.u609280U.	m(Z) 367
	.00363703.	+060116D0+	.06637400.	.07441904.	-072260D0.	80E (1)
	.67356600.	. C75373Dd.	.07666600.	.07779600.	-07877400.	W(Z) 369
*	.3/961100.	. 65031600.	• ng ng ng a	.08130000.	.08173000.	4(Z) 370
	.J&155604/					m(Z) 371
	JATA 227/					w(Z) 372
	. 44400J200 e	.C∪4¥45D0.	.00995200.	.01484 LDO.	.01963200,	w1Z) 373
, •	.6242703.	.uzdElZVG.	.03315800.	.u373taDu.	.041274DU.	w(Z) 374
•	.04562380.	. L 4d 5556 C .	.05186900,	.054962D0,	.06783500.	W(Z) J75
	***C4AF0A*	.co243aDu.	.46517600.	.06721700,	-06906800.	W(Z) 376
*	.u/u/2030.	.01423200.	.0735c3DU.	.07473900.	.07577000.	w(Z) 377
	.67466400,	- 617436b6,	.01001000.	.078612D0.	.07904400.	W(Z) 378
	.u?atotbo/					W(Z) 379
	unta Zdd/					W(Z) 380
	. COUCALOU.	.00468560,	*00477420*	.01393500.	.018446DO.	w(Z) 381
	.02264700.	.02711800.	.03123900.	•J35195D0•	.03897400.	W(Z) 382
	.04252500.	C4556200	-04916100 •	.052159D0.	.05495800.	#(2) 383 #(2) 384
	.05/EE70U.	. C5990zU0.	.0621770u,	-0642U6DU.	.06605800.	W(Z) 385
	.Ju773dUV,	. Cc9254UG,	.0706150J.	.J71829D0. .J75961D0.	.07290200. .07645500.	W(Z) 386
	.07384500. .07885500/	.0146616.	2073366004	#012 301 004	1010455004	w(Z) 387
	1070055007 Dafa 2357					W(Z) 388
	******	- (0440666.	. 308 78603 .	.01311500.	.01737000.	W(Z) 389
	4021529Dv.	.u2::74uC,	.02944600	.u.33253DQ.	.03666100.	W(Z) 390
	. 64436100.	.69356766.	. 4005300	.04955600.	.05227900.	W(Z) 39L
	.05431400.	.05717900.	.03930<00.	.uo137400.	.06321900.	w(Z) 392
	.004903000		.00781503.	.00905800.	.070166DQ.	w(2) 393
	.67114500.	.67241304.	. 0727c4Du	.07341100.	.07395900.	W12) 394
	.0/441000/		_			w(2) 395
	UMTA ZAUZ					w(Z) 396

IA & CFAFF ST	8LK 1	DATA	DATE = 751	78 09/00	/45
* *************	. 40415306.	•0-428200,	.3123640		w(Z) 397
* . 62432480.	• L24 162DJ•	•45\P8000•	-0314690	0, .U3491690.	W(2) 398
* .U30212v0.	.U4135260.	• U4432dDU •	•u471390	004y78350.	W(Z) 399
• -USScC001.	• Ct+ 5 72bu+	.05072000.		006054000.	w(2) 400
• .JUZZEZDU.	· (C) 15900.	.00212000,			W(Z) 401
• .Jods/2Dv.		7326700,	.0709590	uu7155500.	W(Z) 402
4 .4/244104/					w(Z) 403
UMIM 2417					W[Z] 404
#		* 70000000	*00010000*	.0000000.	WIZJ 405
• .00000000.		• 00000000	*0000000*	.00000000.	H121 406
• 00000000		* 10000000.	*00000000	.00000000	W(2) 407
* .dudubdbj.	.360366006	.00000000	********	.00000000.	W(Z) 408
* .66066630.	- 0 6 0 0 6 0 0 0 0	. 696999999	.00000000	.00000000	W(Z) 416
*					W(2) 411
JAIA 242/					W(Z) 412
+ .1126a40J,	-04433200-	. 00020000	.00541000.	.05889700,	W(Z) 413
+ .u51u4u00.	-U44524DC.	.43940400.	4J440500 .	•U30566Du _*	W(Z) 414
* .02724200.	. 4 4 4 5 2 10 4 .	.02193400.	. 01980 > Du .	.01795100.	W(Z) 415
 * *61032989* 	**********	. U13c4aDO.	-J1253600,	.01154700.	#(2) 416
*	• 667474Pr•	. 4.910364.	.,0852630.	.007949DU.	H(2) 417
* .UU742/UU.	.U.ai520	.006524444.	.04612580.	• uJ5764EQ.	W(Z) 418
• .cc5423Du/					#(Z) 419
DATA ZAZZ					W(Z) 420
* •41572300•	.1022206.	.1±14u3uu,	.13473900.	.1161470u.	₩(Z) 421
* .Lag/a2au,	•00/95306•	.07727500.	•00c55bot•	.06056380.	w(4) 422
* acceptance	48 17 20	.04354200,	-134730D0.	.03567100.	W(2) 423
• .03246300.	e bitu43bc.	.02713480.	•05434RD0•	.022987CV.	W(Z) 424
4 audlaseba.		• ul a5oggg •	*01P3A100*	.015845CU.	W(Z) 425
* .u14dutbu.	• 71796500.	• 61709500 •	•01221000 •	.01149800.	W(Z) 426
* +clues/Du/ Cata 244/					WIZ) 427
* .3100160.	. ၁၀ ၁၀ မေ <u>)</u> (၂)	.22905300.	.19793700,	130 50 300	W(Z) 428
* .1475cbDu.	. L = >4 u c D G .		-10004700.	.170203E0. .089444C0.	W(Z) 429
* -U79ac4Du.	. C7162EDu.	-00451000.	•020759D0•	.05293660.	#[Z} 430 #[Z] 431
	. 04465106.		.J.7116Du.	.03421700,	w(Z) 432
* .Jaložedv,			•952332D0•	.02363300.	4(Z) 433
* _u22(500).	.UZU66706,		.01824100.	.01717200.	W(Z) 434
* .Lic15200/				·	W(Z) 435
UNIA Z4=/					#{Z} 436
.dorffon.	.344020Dt.	- e 4400JUU .	.23301380.	.21 7 76000.	#(2) 437
* *!#!#ddn0*	• 1 C 1950Dr •		.12110100.	.llo7140v.	m(2) 438
* «1v4)duDu,	•053752UL,		.07053880.	•00451900•	W(Z) 439
* • • • • • • • • • • • • • • • • • • •	.05797606.		.04d/3100.	.U45139DU.	w(Z) 440
* - 441/4dJU.	-03c7LuDL.		.03344000	و لا حال خال عالم	W(Z) 441
* .L2y2_4D0, * .L2\4c60U/	•0273¢y0v•	.u257000u.	.02410800.	.022759WO.	W(Z) 442
PATA 446/					W(Z) 443
# .4/0520Vu.	. 4664 7406.	•354751 Du •	.30312400.	- Jo Ibo Iwo	W(Z) 444
• .23648800.	.20200206		.1570d7E0.	.20150300, .14199580,	w(Z) 445 w(Z) 446
• .12/2C2D0.	.11446000		.434164	.00529BCU.	w(Z) 447
*7/85133.	.07128300.		•46956359•	•43560 l Du •	#[2] 44d
• .db152100.	· L+70C4Lu.		.04142000.	•079090f0•	e(2) 449
* .J7019600.	. 03362606.		•45AA0009•	.02823160.	W(2) 450
* .vcov3oDu/		_			w(Z) 451
DATA 247/					w(Z) 452

IN C TEAFT 51	BLK	DATA	DATE = 751	7 4	09/06/45
* .53571300.	.45966500.	.396852D0.	.34464500.	.30098900.	w(Z) 453
* _26426HDU.	.0903SLES.	.206787DU+	·18420000·	.16479300	
* .14d036D0,	.13354LDG.	.120838D9.	.109759DQ.	. LUGOZ6DU.	H(Z) 455
* .09144300.	.U#3845DO.	.07709600.	.07108100.	+065701D0	W(Z) 456
• .0608760 0 .	.US65J4D0.	.052617D0.	.04907300.	•04585900	H(Z) 457
• .04294aD0.	.04027100.	.03783600,	.03560700.	.033561D0	
• .G316±0DQ/	-			•	¥(Z) 459
DATA Z4#/					M(Z) 460
• .57au42DQ.	.457744DC.	.43244200.	.37766800.	.33153500	
* .29243200.	.254136DC.	.23064600.	.206155D0.	18500500	
• .ióbbc0D0,	.150bE1DO.	.1367u600.	.124435D0.	-11362000	
* .184954Du. * .869748Du.	.05556300.	.08800100.	.081245D0.	+075 19000	
* .049417Du.	.046364DC.	.043008D0.	-056391DJ.	.05274100.	
* .U30577D0/	**********	*********	.041G64D0,	.03872800.	
UATA 249/					W(Z) 467 W(Z) 468
• •••••••	.52293200,	.457569DQ.	-402194D0.	.35508200.	
* .31482ED0.	.28029000.	.250532D0.	.22476900	.20242900	•
• .182932Du.	.165868DC.	.15087700.	.13706100.	.12597 LDO.	
+ .115654DU.	.106355DC.	.00501860.	.09071000.	.084068E0	
* *07808±00.	.4726ECDG.	.06778500.	.06334200.	.059298DO.	
+ .055ulca0.	.0 5223 ಕು ೯,	. C4915000,	.04631500.	.04370800,	
\0030E140. *	•	-		-	W(Z) 475
CATA ZEU/					W(Z) 476
# .61014cDU.	.53006700.	.47277300.	.418491DO.	-37181300-	H(2) 477
* .331544Du,	.25665204,	.26642700,	.24005700.	.21700400.	
• .1967a3Du.	.178990Dt.	.16328100.	-14937000.	-137012DD.	w(Z) 479
12=00200.	.116164DL,	.10734800.	.09942700.	.09229100,	W(Z) 480
* .Q45845D0:	.0800090;.	.074712D0,	.069894D0.	•005500Du	W(2) 481
.00148aD0. ★	.05781100.	.054439D0.	.05133900.	.04848500,	w(2) 462
# .U458:1DQ/					W(Z) 483
UATA ZEL/		. =			W(Z) 484
* a00715d00,	.53855600.	47899100.	.4272250U,	.38210500.	
* .3428/200.	.36853000.	.27844500.	.25202400,	.228759D0,	
+ =20821900+	-15003600.	.17389600.	-15953100.	.146712DQ.	
* .13524200. * .49255800.	.12495400.	.11570200.	-107.36100-	.099824D0.	
* .Ju7u24Du.	.00308000.	.08116200.	.07602100.	.U7132400.	
* -35015700/	140300001	**********	4030110001	***********	w(2) 491
DATA 452/					W(2) 492
+ .5937c1D0.	.532009D0.	.47743900.	.42927500.	.386777DO.	
* .44920000.	.41612800,	.28081500.	-200847D0.	-237800D0.	
* .2173CEDU.	.149046D0.	.18274200.	.16815100.	.155066D0.	W(Z) 495
+ .14330500,	-13271100,	.12J147DU.	-114495DQ.	.10665000.	
. 04952300.	.05304500.	.08711600.	.08170600.	.07675300.	m(2) 497
* .u722uauu,	.04841100.	.00418000.	•06063900•	.05736300.	w(Z) 498
\u011E66U. *				•	W(Z) 499
DATA 2537					W(Z) 500
+ .57239700.	.51446300.	.46948BDJ.	.42566700.	.38641200,	W(Z) 501.
* .35129 9 00,	.31991800.	.29185100.	.26675700.	.24429500.	W(Z) 502
* .22418800,	.2Ct1 CEDU.	.18987800.	.17527100.	+1021J0D0s	Y(Z) 503
* .150205Du.	.1J9441D0.	.1290d4DU.	.12082£D0.	.112760DO.	W(Z) 504
# .lu541100.		.09256200,	.08637900.	.081773DQ,	w(Z) 505
# .077024D0.	.07265100.	.u68617D),	.06489000.	-061440D0.	W(Z) 506
+ -05d24300/					W(Z) 507
UATA ZE4/					W(Z) 508

IV G LEVEL 21	dLK	DATA	UATE = 751	76	09/06/45
# .54545EJJ.	.45521000.	4456555D04	-41749100-	.38190800	W(Z) 509
* .349011Du.	.42446800.	29392700.	.27004000.	.248402DJ	
* .22d9c7Du.	.21134300.	.19539803.	.1809570u.	•16786300 e	w(Z) 511
* .15597 5 00,	.145167U¢.	.135326D¢.	•12o353Du•	.11815sCO:	W(Z) 512
* .llucc200.	.14374506.	.097495Du,	.0917U6D0.	. UU6376CO	■(2) 513
• .U81467UU.	. 676533DG.	.0/274200.	.06686380.	•Qa52a6ÇU,	₩(Z) 514
* .00192000\					¥(Z) 515
DATA Zob/					W(Z) 516
* 00F17q1e* *	.47653506.	.44600500.	.40562300,	.37411000	
# .3448EdDu,	.31362206,	.29345300.	.27101500.	.250549CU	
+ .2Jte57DC,	.214902000	-199416Du.	+1 d52y900 ·	.17242JCO	
# .1006tdDJ. # .11528cD0.	•145527DG•	.14010300.	.13110600.	12255800	
* .00523800,	.108325D6,	.1u1y1y00.	.09601500.	#490567D0	
\u0033500,	1001013041	. 0. 633, 601	• U72553DU •	-008834D0;	
DATA 450/					W(Z) 523 E(Z) 524
,0U/\$\$Lup. *	.4517c3bG.	.42107600.	.JyloosDu.	.363828D0,	
* .33772000.	.31339700.	.29U847DU.	-270ul6Du	.250823DO	
* -23417100.	.21695404.	-202667DU.	.1804u3Du.	.17586200	
* 10454900,	.15377304,	.14405400.	.13511300.	.12688300	
* .11929800.	.L12302D0,	.105842DU.	.09987000.	.094343D0.	
# .Ca92220J.	.084472DG.	.00100100.	.07596000.	-072142D0	
# .Ocase5D0/					W(Z) 531
UATA ZE7/					#{Z} 5J2
# #4512440U.	.426168DC.	.4008J7D0.	.37591100.	.J518u3D0.	W(Z) 533
* .328777 ₀₀ ,	.3C&\$\$ODU.	.2do517D0.	.207J78D0.	.24955600.	w(Z) 534
* •5330C 9 D0 *	.21707aUO,	.203494DU.	-190384DU·	.17827500,	w(Z) 535
# .167092Du.	-1567¢50c,	.14722600.	.138412D0.	•130202D0,	¥(Z) 536
# .122723Du:	-115744D0,	.109277DJ,	.1u328uDu.	.U977L300,	
# .U9254100.	.OE7732DU.	.08325400.	.079082DJ.	.J7519100.	
* .07155800/					m(2) 539
DATA ZEB/					W(Z) 540
* .42038a.00.	.40074300.	+38CI6IDO+	.35931300.	.338070001	
* .318584Du.	.299261D0.	.260846D0.	.26341800	.24701200.	
.0008891F7* +	.217253Du,	.20384700.	.19136600.	.17976200.	
* .12559000,	.[58969DQ.	.14967400.	.14104500.	.13303300.	
* .095494000	-05066GDQ.	.035414300.	.udl925000.	.07798200	
+ .07424300/	,,	*****	140125001	1471702001	W(Z) 547
DATA 459/					u.(Z) 546
* .J91291Du.	.37521400.	.35972100.	.342479D0.	.324985CJ.	
+ .307609DJ.	.25001300.	-274180Du -	.25843100.	. 243439DJ	• - •
* .22924400.	.215827D¢.	-20327200.	-191471DG .	.180425EU.	W(4) 551
+ .17064900+	·160457DQ.	.151458D0.	-14306300.	·135234D0	w(Z) 552
* .1275310J.	.12111HDG.	.114761CO.	.108827D0,	.10328500.	w(Z) 553
• .0981C7Du.	.0932¢500.	.086735D0.	.04449300,	.u4051900:	w(Z) 554
+ _L7679400/					W(Z) 555
VOSS ATAU					w(Z) 55t
* .36442700.	.35306600.	.34000400.	.325873D0,	-31116100.	
* .29624UUU,	.281392DQ.	.26a82JD0,	.252od100 ·	.23900700.	¥{Z} 558
* -22604600 ,	.21365600,	.20151400.	.190821D0.	-100Jo70U,	
* .170534D0,	-16130000.	.15263700.	-144516D0-	.136908Do.	M(Z) 560
* .12578100. * .10037800.	.12310800.	.11685800.	-1110030J.	• 105519D0 •	
* .10037800. * .0790t50u/	. 095558800.	-091037D0.	_GB67940Q.	.08280900,	W(Z) 562
DATA 261/					#(Z) 563
DAIR 211/					W(Z) 564

IA C FEAFT 51	BLK	DATA	DATE = 751	78	09/06/45
* .340026Du.	.3315e3DG.	.32133200.	.30983100.	-29752900	₩(Z) 505
# .264786Du.	.271681Da.	.259031D0.	.246J96D0.	.234096DU	w(Z) 566
* .22221300,	.2108C5D0.	. 149904D0.	.109524D0.	-1796870U	w(Z) 567
# a170371DU.	-1-1572DC.	·15327400·	.14545700.	.138100DU	w(Z) 568
+ .131180D0.	.124674DG.	-118558D0·	.112810D0.	.1074U8DU	w(2) 569
10232900.	.097554D0.	-093062D0-	.088637D0.	-08485900	w(2) 570
• •08111300/					W(Z) 571
DATA ZG2/					¥(Z) 572
♦ .318073D0 ,	.31186600.	.303894Du.	.294574D0.	.28432700	w(Z) 573
• .2734E2D0,	.2623 CŁD Q.	.25101600.	·239772D0·	-22670300	w(Z) 574
* .217904DU.	.2C7442DU.	-1973oóDJ.	-L8770500.	.178478DU.	w(Z) 575
* .16969100,	.161343Du.	. 639429D0.	•145930DU•	+L3885500	w(Z) 576-
-132144D0,	.125849DU,	-11989100.	·11427200 ·	-108973D0	w(Z) 577
* .10357780.	-09926500.	.094822D0.	.09063180.	. UB6677D0	w(2) 578
+ -082544047					W(Z) 579
DATA ZE3/					W(Z) 580
+ .298408D0 ;	.29J982D0,	.287771Du.	.280232D0.	.271710DQ	w(2) 581
-26249900-	.25284400,	.24294700.	.23296800.	-22303700	w{Z} 582
• .21J253Du,	.203092DC.	.194410DQ.	.185446D0.	.176827DQ	w(Z) 583
* .lo85c9D0.	-140680DD.	.15310100.	.146009D0.	-1392170u	w(Z) 584
.13277306,	.12666700,	.120892D0.	-11541300.	•110236Cu	w(Z) 585
• .10=33900•	.LGQ7CSDu.	.096330D0.	.09218900.	.06827300	w(Z) 586
* .08456800/					W(Z) 587
DATA Z64/					w(Z) 588
- 28102600.	.277755DU+	.27296000.	.26686500.	-25977500	#(2) 589
* .251 ~ 5300.	.24J617D0.	.23445200,	.22611100.	.21721900	• ₩(Z) 590
+ .20837tDu.	-166PE0Dn+	.19113300.	*1858*ADA*	.17481450	W(Z) 591
<u>*</u> -16707800.	.ltvo46Di.	.15252688,	.14572100.	.13922900	#(Z) 592
♦ .133U4€DO.	.127161DC.	.12156900.	.116250DQ,	.11121800,	w(Z) 593
• .luo+3oD0.	-1019C1Dc.	.09760100.	.09352300,	.08965800	W(Z) 594
* .0859 92 00/					W(Z) 595
CATA 265/					w(Z) 596
* .20552200 ,	.26J2Q1DC.	.259435DU.	.254476DQ.	•248566DO:	w(Z) 597
4 -24191400.	.0 GP17465.	.22712900.	.219302DQ.	.21134900	M(2) 598
.ud83t605. *	.195438Do.	.187620DQ.	.179965D0.	.17251000	W(Z) 599
+ -16528100 .	.15829900.	.15157600.	.145120D0,	* 138A3200	w(Z) 600
* -133 61500 .	•127363DG•	.121972DO.	.11683400.	-11194200	
* .10728600.	.lu2a58D0.	.098648D0.	. 09464600,	-09084200	
* .08722700/					w(Z) 603
DATA_ZEG/	•				W(Z) 604
• .251723DO.	-250050DC.	•247U92D0•	.243042D0,	.23809200.	·•
* -23242000.	.2261 500 C.	.21954600,	.21261400,	.20550400.	·
• .19830700,	.191099DC.	.18394300.	.176884D0.	.16997700,	
4 .163237D0.	.l 56692Do.	.15035900.	.14424900.	.138368Du	
1327∠0D0.	-127305D0.	-12212100.	.117164D0.	.112428CO,	
10790900.	I ¢32 240 0 • "	-09948700.	*095570D0 <u>*</u>	-041839CD	
• .08828300/					w(Z) 611
DATA 267/	. 22utožne	23683800	9 2 9 6 4 5 6 4	2227776	W(Z) 612
# .239403D0.	.23818700.	.23583800.	.23250400.	-22833700	
* 22J462D0,	.21807700.	-21224700.	.20610300.	. 19974400	
* .193255DQ,	-160767DD-	.18016JDQ.	.173670D0.	.167270CO.	
* .100996D0. * .132191D0.	-154872D0-	148918D0.	.14314700.	13756900	
* .13219100,	-127015DG-	.12204200.	.11727100.	.112699Cu,	
+ .U832200; + .U89170D0/	•1 C41 36D C .	-100133D0,	.09630900.	.09265700.	
DATA 208/					■(Z) 619
DAIA 2007					W(Z) 620

IN G TEART 51	BLK	DAŤA	DATE = 751	78	09/06/45
• .22835£0¢•	.22745±00.	.22556900.	.2228U0D0.	.219268DO	w(Z) 421
• •21509300.	.21 G3 6 7D 0 .	.20525800.	.199804DQ,	-19411100	
• .19952990•	.162311DG.	.17632800,	. L7035700.	.104438D0.	W(2) 623
_# .l586Q4DJ.	.lt2862DC.	.14729200.	-14185100.	.136571D0	¥{Z} 624
* .131459DQ.	.12652200,	.121762D0.	-117L80D0.	.11277500.	w(2) 625
+ · 10854600,	.1 G44 £90 L.	.1006u1DJ.	.096876D0,	.09331u0u,	¥(Z) 626
+ .08985800/					W(Z) 627
DATA 264/	0133300				W(Z) 628
• .2183990J. • .207232D0.	.2177220	.216181DO.	.213858D0.	.21084300,	: -: - - :
-016d35400	.203119DQ.	.198594D0.	•193741D0 •	18863800	-
• .15609900.	.150758DC.	-14551800.	-16699000.	14151900	
* .LJUSE3DQ,	.12545106.	.12130300.	.11691100.	.1354J300.	
* .10859700.	-104674DU-	.10090500.	-J97284UO.	.09381000	• •
# .090479D0/				10,550,550,	¥(2) 635
DATA 270/			•		¥12) 636
.20y377Da.	.208854D0.	.20757700.	.20560700.	. 20301480.	
• .19987300.	.15626200.	.19225bUO,	·18792700.	-18334400	
• .178548DQ.	.17Ja54D0.	.168651DU.	.163603DQ.	.15854700,	
• .153£15DU.	.142534D0.	.14302500.	.13880700.	.134094CO.	
+ .12949800.	.12502700.	.12668880.	-116484D0.	· 112419D0,	■(Z) 641
* .10849300,	•104707VG•	.101058CU.	.097546DQ.	.09416800	W(2) 642
* .09C9E1D0/					E46 (1)
DATA 271/					W(Z) 044
• .2011£700,	.20074200.	.19966900.	-19798000.	.19573200.	
* .192584DQ;	.165798DG.	.186234D0.	.182368DQ.	.17824300.	
* .17391±D0, * .15086000,	.109445DG.	•164dobD0•	*16022300 •	-15555100-	
* .12631700.	-124071DG.	+14164000,	-13711300	.132667D0,	-
* .10824900.	-1 C46 OGLO.	.10107600.	.11591900. .09767400.	.11202300, .09439500,	
\u0452140. +		**********	1037074001	.024393001	w(Z) 650 w(Z) 651
DATA Z72/					W(Z) 652
	-15325200,	.15237600.	.190915Du.	-18695100-	
• .186532DJ,	.14374506.	.18053480.	.17706100.	-173340D0.	
• •los41qDû.	.16533500.	.161145DQ.	.15687200.	.15255300.	
• .14d£17DU.	.14388800.	.13428800.	.135J35D0.	-131146D0-	w(Z) 056
• +12/u31Du-	.1236L3D0,	*11A00RD9*	411523300 .	·11150300.	* W(Z) 657
* .107e8luu.	.16437006.	.10066900.	.09768400.	.09450200.	W(Z) 658
• -U91434D0/					w(Z) 659
DATA 273/					4 (2) 660
* .13070400.	.18042100.	.10563000.	.18435400.	.10262600,	
* .1d048400. * .1d507200.	.17797000.	-17512800.	.172003DO.	.16863700.	
* .14554500.	.161344Dc.	.157502D0.	.153567DD.	14957200.	
* .12566000.	.121 # 40DG.	.11809900.	-11444200.	.129548DQ,	W(2) 664
* .107403Du.	·1U4027D6.	.1 c075100.	.09757500.	.094499D0.	₩(ZJ 665 ₩(Z) 666
+ .09152300/				107-17750,	W(Z) 667
DATA Z74/					w(2) 668
* .1804C2D0,	.10006100.	4179369D0.	·178245C0.	.17671500.	W(Z) 669
* .17480HDU.	.1725EGDU.	.170006D0.	.1671840U.	.164132DQ.	W(Z) 670
• .loueeedu.	-1 = 7 4 E CD + .	-15394#PO.	.15032000.	.14062300,	₩(Z) 671
* a142682DQ.	+13412000.	.135357D0.	.13160900,	-12789200,	H(Z) 672
12421900.	-12uo00D0.	·11704500,	11720000.	•110153Du•	■(2) 673
• .106827D0.	•1u3586D0•	.lu0433Du.	.09736980,	.09439600.	W(Z) 674
+ .09151JD0/ Umfa 275/					W(Z) 675
					W(Z) 676

IN & FEAT	21	BLK	DATA	DATE = 751	76	09/06/45
÷	.17436200.	.174152DC.	.173542D0.	.17254500.	-17118100	₩(Z) 677
	-16947500.	.16745500.	.16515100.	.162596DO.	-15982100	. #(Z) 678
	.156858DO.	-15373600.	.150490D0.	-147141DO-	.14371700	w(Z) 479
	.140239D0.	-13673100.	-1332Q9D0.	·12969100·	+12619200	w(Z) 660
•	.122724D0,	-11929600.	-112513DO·	+11260200.	.10934900	, #{Z} 681
•	.106166D0.	.1 C3057D0.	-100026D0.	.09707300.	.094202D0	<u> </u>
	-07141300/					E86 (I)U
	DATA Z76/					W(Z) 684
•	.168830D0.	.168645D0.	.168102D0.	.16721200.	-16599404	w(Z) 645
	.164456DU.	.162633D0,	.160548D0.	-158227DO.	- L5569800	. <u>u(Z) 686</u>
•	.15258800.	.15012400.	-14713200.	.14403800.	.14086200	. ¥(Z) 687
	.137626D0.	-134354D0.	431058D0.	<u>-12775500.</u>	.124460D0	. <u> </u>
	-12118500.	.117940DG.	-11473500.	·111576D0 ·	-108474D0	, ¥(Z) 689
	.10543100.	-10245100-	.099534D0.	.096696004	-093927D0	w(Z) 690
	.09123000/					4(Z) 691
	DATA 277/ _					W(Z) 692
	.16366200.	.163498D0,	-16301100,	.162211DO.	-161111100	. w(Z) 693
	•159725UQ+	.1 £8075D0,	.15618100	-154066D0.	<u>.15175800</u>	. <u> </u>
	.14927100+	.146637DG,	.143878DO.	.14101400.	-13896709	. ¥(Z) 695
	·135056D0+	.13199900.	.12891300.	.125812D0.	.12270900	. W(Z) 696
Ť	.119b17D0.	-11654500.	.113503D0.	-110500DO+	.10754000	. ¥(Z) 697
<u>.</u>	.1046JIDQ.	.16177700.	.09898100.	.09624700.	.09357780	. Y(Z) 694
•	.09057300/					W(Z) 699
	DATA 276/					W(Z) 700
*	-1588 -100 +	.158673D0.	.15823500.	·15751300·	.15651600	. ×(Z) 701
	.15526000.	.153760DC.	-152034D0+	-15010200.	-14798500	
	.14570300.	.14327700.	.140727D0.	.138074D0,	.135336DQ	. W(Z) 703
<u>*</u>	.132530DQ.	.129674DQ.	-126782D0.	•123869DO•	.12094700	
•	.118027DO.	.11512000.	.11223400.	.10937700.	.10655600	
*	.103777DQ.	• î ĉ î ĉ 44 <u>00 •</u>	.09836200	.09573400.	-08314500	
*	.QYV649DQ/					W(Z) 707
	DATA Z79/					V(Z) 708
	.15427JDu.	.154140DQ.	.15374300,	. 1 5308600 .	-152183D0	-
_*	151C4CDO+	.14967200.	-148094D0,	.14632400.	-144380 DO	
	.142279DQ.	.14003 9 00.	.13764000.	.13521800.	. 1 3267 LDO	
.*	.L3qu54Du.	.12736400.	.12467300	.121935DO.	-11918200	<u> </u>
*	.116425D0.	-11367300.	.11093500.	.108218D0.	.10553000	
*	-102875D0-	-1 COS 6 0D 0 +	.09768800.	.09516JD0,	-09 2688DD	
*	•690265D0/	_		•		W(Z) 715
	DATA ZEO/					V(Z) 716
*	.149992Du.	.14987100.	•14951 QDO •	.14891300,	.14808800	
	-147044DU	.14575200.	.144346D0.	.14272100,	14093100	
	.13899300.	.13692200.	.134735D0.	.13244800.	-13007600	
	.127633D0.	• 1 251 33D C •	.122591Da.	12001600	+1174220G	
	-11481700.	.11221200.	.L09614D0+	.10703150.	.10446900	
	•101 S3500 •	099433D0.	-096968D0.	.044543D0.	.09216200	
	.08982600/					¥(Z) 723
	END					W(Z) 724

```
IV G LEVEL 21
                                                   VATE = 75178
                                 MAIN
                                                                          09/11/18
           DIMENSION HEADS(20)
           DIMENSION HISTOR(36) . ENSTER(30) . TAUSTH(30)
           UIMENSIUN ANGLZEDD).HT(2500).PLCTA(2500).PLCTY(2500).
          *XPLI(4), YPLT(4).dUFF(1000)
           NCASE= 0
           I = TART= L
           STMAX=-100C.
           ** 17E1c. 81527)
     01927 FURMAT(181///)
           CALL PLUTS(SUFF, 4000,10)
           LALL PLUT(C. . - 12. . 3)
           LALL PLUT(0..-11.5 .23)
     C
     L
           READ IN CURTRUE PARAMETERS
     88366 REAU(20.END=4564)SLIT.PLTNIN.FLTPAX.DELPLT.DLAP.YHGT.NLINES.ENO.
          *TL.TA.EL, [PLOT1.]PLOT2
          F.LIM2.APRIME
          ♥ • FAU
          *. FEADS
           THAF=(-1)++L1X2
           READ IN MAYENUMBER. TRANSMISSION PAIRS. CONVERT WAVENUMBERS
    C
           TO WAVELENGTH AND DETERMINE MAXIPUR THANSMISSIGN
           DO 837 ICARD=1.NLINES
           HEAD( 20 JANG( ECARD) .HT( ICARD)
           IF(HT(ICARD).LT.O.)HT(ILAHD)=U.
           STMAX=AMAX1(STMAX, HT(1CARD))
           ANG(ICARU)=(1.E+U8)/ANG(ICARU)
       837 CONTINUE
           NCASE=NCASE+1
           II=NLINES
           PL TH [N=ANG( 1)
           PLTMAX=ANGLELS
           wilder Zo + DL AM
           PLIMIN=PLIMIN-2. ** 105
           PLTMAX=PLTMAX+2.+51DE
           DELTA=STMAX/S.
        35 CUNTINUE
    C
           PRODUCE THE ZERO SLIT MINTH PLOTS WHEN I PLOTI=1
           IF(IPLOT1.EQ.0)GU TU 65001
           TEST=(PLTMAX-PLTMIN)/UELPLT
           ILG= TEST
           ILG=ILG+2
           HLG= ILG
           JPLT(J)=PLIMIN
           XPLT(4)=OELPLT
           YPLT(3)=0.
           YPLT( 4 )=DELTA
           BOUZE=YPLT(4)+1C.
           LALL AXIS( ..... *A *,-1 . RLG. 0. . XPLT(3) . XPLT(4) . 10.)
           YPLT(1)=6.
           DG 50 1=1.11
```

```
IN G LEVEL 21
                                                   DATE = 75176
                                                                         09/11/18
                                MAIN
           RLAM=ANG( I )
           IF (PLTMIN-GT-ALAM-DR-PLTMAX-LT-RLAM) GG TO 50
           YPLT(2)=ANINI(HT(1), GCCZE)
           XPLT(1)=RLAM
           XMLT(2)=XPLT(1)
           CALL LINE(XPLT.YPLT.2.1.0.1)
        SU CONTINUE
           CALL PLOT(REG+2..0..-3)
     66001 CONTINUE
     c
           PRODUCE THE ACTUAL SPECTRA PLET
           ISAVE# 1
           KKQUNT=1
           BELLAME (PL THAX-PLTMIN) /2459
           PP 1=PL TH IN-DELLAR
       441 LUGK= | SAVE-1
           IF(LOOK.LE.G)LUCK=1
     c
           CHOUSE A PUINT ALONG THE FLCT AXIS AND SET AN INTERVAL OF WIOTH
           EQUAL TO THE SLIT WIDTH ON EITHER SICE
     L
     c
           P1=PPL+KKULNT+DELLAM
           P2=P1+w10L
           XLAM={PI+P2}+.5
           KUUNT=1
           THIS LUGP DETERMINES WHICH LINES LIE IN THE CHOSEN INTERVAL
     c
           DO 5565 I=LOCK.II
           IF (ANG(1).GT.P2)GU TO 992
           IF(ANGII).LT.PL)GU 10 5569
           IMINUUNT .EC. 1) ISAVE=1
           KUUNT=KUUNI+1
      SSOS CUNTINCE
       SEZ VUPTSEKOLNI-1
           THE FEELUNING STATEMENTS SIMULATE THE TRIANGULAR SLIT FUNCTION
     L
           OF SUMMING THE CONTRIBUTIONS OF ALL THE LINES IN THE INTERVAL
           LICHUPTURE URDIGE IL SS741
           SUMMER = O .
           [PNI=1SAVL+NUPTS-1
           DE 99347 JEE=15AVE+IPNI
           TREADURATE JUL ) # (UL AM-AB = ( XLAM-ARG ( JUL ) ) } / DLAM
     99347 SUMMER = SUMMER+AMARICO. THIAUDI
           PLUTX(KKJUNT)=XLAM
           PLUTY(NKOUNT)=SUMMER
           KKGUNT-KKGENT+1
           IF(KALLNT-EG-2561)GG TU 77
           GG TU SSI
     99741 PLOTXIKKGUNT)=XLAM
           PEUTYLKKUUNT I=G.
           KKUUNT=KKUUNT+1
           IF (ANGILL) .LT.PIJGU TC //
           IF (KKULMT-EL-25CI)GC TO 77
```

```
DATE = 75178
IV G LEVEL 21
                               MAIN
                                                                        09/11/16
           GU TO 991
        77 II=KKOUNT-L
           N=1I
           TTRAF= .5+TRAF +DLAM
           DC 68 [-1.A
        88 PLOTX(1)=PLCTX(1)+TTRAF
           TEST=(PLTMAX-PLTM(A)/DELPLT
           ILG=TEST
           IL G= 1LG+2
           ALG= ILG
     C
          OFTERMINE THE MAXIMUM VALUE OF THE SPECTRAL CURVE AND IF THIS
     c
           IS THE FIRST CURVE COMPUTED. THE Y AXIS SCALE FACTOR IS CALCULATED
     c
           SPMAX=- 100 .
           DD 800 I=1.N
           LF(PLGTY(I).LT.SPMAX)GC TG 800
           SPMAX=PLOTY(I)
       800 CONTINUE
           IFIISTART-EU-L)DELYP=SPHAX/YHGT
       BUL PLOTY(N+1)=0.
           PLOTY(N+2)=DELYP
           PLUTA(N+1)=PLIMIN
           PLCTX(N+2)=DELPLT
           VALMAX=10. #PLOTY(N+2)
           DO 65395 MUNDAY=1.N
           IF (PLOTY (MONDAY) .GT. VAL PAX) FLCTY (MONDAY) AVAL WAX
     65395 CUNTINLE
           NP2=N+2
           WHITE(40)NP2.(PLOTX(L).PLCTY(L).L=1.NP2)
     C
           PRUDUCE A SEPARATE PLOT FOR EACH VALUE OF ENG IF 19LOT2=1
     C
           IF(IPLOT2.EQ.01GO TG 66CC2
           CALL AXIS(U., U., A4,-1, FLG. O., PLOTX(N+1), PLOTX(N+2), 10, )
           CALL SYMBOL(.5,5.0.1,'NO='.0.,3)
           CALL PLIFLI(-0.,-0.,-0.,EAC.G.,3)
           CALL LINE(PLOTX.PLDIY.N.1.0.1)
           CALL PLOT(RLG+2..0.C.-3)
    66002 CUNTINCE
           ISTART=J
           TAUSTRINCASE J=TAU
           HTSTORINGASE 1=SPMA X/DELYP
           ENSTOR (NCASE)=ENO
           GO TO Ed368
      4994 CONTINUE
    c
           PRODUCE THE FINAL SPECTRA FLCT
    c
           LALL AXIS(C..O..'WAVELENGTH' .- 10.RLG.Q., PLTMIN.DELPLT.10.)
           CALL SYMBUL(.5.5.9..1. HEADS, 0..80)
           CALL SYMBOL( .5.5.5.1. 'TE= .0.,3)
           CALL NUMBER(-0.,-0.,-0.,TE.O.,-1)
           CALL SYMBUL(.5,5.3,.1,474=1,0.,3)
           LALL NUMBER(-U. .- U. .- C. . TA . C. .- 1)
           CALL SYMBGL( -5.5.1.... *L= *,0.,3)
```

```
IV G LEVEL 21
                                       MAIN
                                                             DATE = 75176
                                                                                         09/11/18
              CALL NUMBER(-G. .- Q. :-G. .EL .G. .1)
              CALL SYMBOL( .5,6.9,.1,3HA 42.0.,3)
             CALL NUMBER(-0..-w.s-0..APRIME.0..5).

CALL SYMBOL(-.465715.6.7.1.* NO MAX HY TAU*.0..19).

CALL SYMBOL(-.265715.8.55..1.* (1N)*.0..13)
              UD 78 I=1.NCASE
             Yb=8.0-.15+1
CALL PLTFLT(-.385715.YW..1.ENSTOR(1).0.,2)
              CALL NUMBER( .42914.YM...I.HTSTCR(1),0..2)
             CALL NUMBER(1.028578. ** .. 1. TALSTR(1) .0. .3)
         76 CONTINUE
             ENDFILE 40
             REWIND 40
        150 READ(40.END=4995)NP2, (PLOTX(L), PLOTY(L), L=1.NP2)
              N=NP2-2
             CALL LINE(PLOTX.PLOTY.N.1.0.1)
GO TO 150
       4995 CALL PLOT(RLG+2..0..-3)
4997 CALL PLOT(0.0.669)
              STOP
              END
```

NOMENCLATURE

$\mathbf{\overline{T}_{j}}$	Transmission of spectral line (designated j) through a medium, intensity units
ν_{i} , ν_{i}	Wavenumber of the ith or jth spectral line, cm ⁻¹
$\mathfrak{r}_{\mathbf{\nu_{j}}}$	Intensity of source spectral line at wavenumber, $\nu_{\mathbf{j}}$, intensity units
$k_{ u_j}$, $k_{ u_i}$	Absorption coefficient for the spectral line at wavenumber, ν_{j} or ν_{i} , cm ⁻¹
l	Absorption path length, cm
ν	Wavenumber, cm ⁻¹
Ι°ν΄j	Intensity of source spectral line at center wavenumber, ν_{i}^{o} , intensity units
$\nu_{\mathbf{i}}^{\bullet}$, $\nu_{\mathbf{j}}^{\bullet}$	Center wavenumber of the ith or jth spectral line, cm ⁻¹
$(\Delta_s \nu_i)_D$, $(\Delta_s \nu_j)_D$	Doppler width at half maximum intensity of the ith or jth source spectral line, cm ⁻¹
K	Boltzmann's constant, 0.6952 cm ⁻¹ K ⁻¹
$\mathtt{T}_{\mathtt{S}}$	Temperature of gas in light source, K
M_s , M_a	Mass of molecules in light source and absorber, respectively, gm
С	Velocity of light, 3×10^{10} cm/sec
$\mathbf{k}_{ u_{\mathbf{j}}^{\mathbf{o}}}$, $\mathbf{k}_{ u_{\mathbf{i}}^{\mathbf{o}}}$	Absorption coefficient at center wavenumber, $\nu_{\mathbf{j}}^{\bullet}$ or $\nu_{\mathbf{i}}^{\circ}$, cm ⁻¹
a <i>'</i>	Spectral line broadening parameter (ratio of collisional to Doppler line widths)
У	Dummy variable of integration
ω	Doppler frequency function, $\frac{2(\nu_j - \nu_j)}{(\Delta_a \nu_j)_D} \sqrt{\ln 2}$
$(\Delta_a^{\nu}_j)_{D}$, $(\Delta_a^{\nu}_i)_{D}$	Doppler width at half maximum absorption coefficient, $\mathbf{k}_{\nu_j^{\bullet}}$, of the ith or jth absorption line, cm $^{-1}$

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$(\Delta_a^{\nu}_j)_L$	Lorentz width at half maximum intensity of the jth absorbing spectral line, cm ⁻¹
z_L	Effective collision frequency for spectral line broadening, \sec^{-1}
$\mathtt{T_a}$	Temperature of absorber gas, *K
$\overline{\mathtt{T}}_{\!\Delta u}$.	Transmission over the frequency interval $\Delta \nu$ through a medium, intensity units
${}^{t}\!$	Transmissivity over the frequency interval $\Delta \nu$; ratio of incident to transmitted intensity
$\Delta \nu$	Wavenumber increment, cm ⁻¹
е	Electronic charge, 4.80×10^{-10} esu
m	Electron mass, 9.11 x 10 ⁻²⁸ gm
N _J ″	Number density of molecules in the lower state, J", ${\rm cm}^{-3}$
$^{\mathrm{f}}\mathrm{_{J}^{'}\mathrm{_{J}}^{''}}$	Oscillator strength for the transition from the upper state J^{\prime} to the lower state $J^{\prime\prime}$
h	Planck's constant, 6.625×10^{-27} erg sec
$\mathbf{B_{v_o}}$	Rotational constant for the vth vibrational state (ground state, $v = 0$), cm^{-1}
J"	Rotational quantum number for the lower state
F(J ″)	Rotational energy of the J" th rotational state, cm $^{-1}$
N_{o}	Total number density of molecules, cm ⁻³
f _{v′v″}	Band oscillator strength for the $v' \longleftrightarrow v''$ transition
ν J´J´´	Wavenumber of the line corresponding to the transition $v'J' \rightarrow v''J''$, cm ⁻¹
ν , ν ,	Wavenumber at the bandhead of the (v', v") band, cm^{-1}
⁸ J″J′	Rotational strength, or Hönl-London factor, for the $v'J'-v''J''$ vibrational - rotational transition
s	Total electron spin quantum number
N _l	Number density of molecules ℓ other than the absorbing molecule in the absorbing medium, cm ⁻³

σ_{ℓ}^{2}	Effective collisional cross section for the broadening process by the ℓ th type molecule, cm ²
N/l	Mass of the 4th type molecule, gm
λο j	Wavelength at line center of the jth spectral line, cm
p _l	Partial pressure of the 1th type molecule, torr
$ ilde{ ext{M}_{ ext{f}}}$	Mass of the fth type molecule, gm
$\mathbf{c}_{\mathbf{j}}$	Line broadening constant for the jth spectral line, $K/torr\ or\ K/atm$
С	Average line broadening constant for all spectral lines in a given band, K/atm